

Station 3: Solving Quadratic Equations

Part A: Complete each problem by FACTORING.

1. Find the real zeros of $f(x) = 2x^2 + 15x + 18$.

$$0 = 2x^2 + 15x + 18$$

$$0 = (2x + 3)(x + 6)$$

$$x = -\frac{3}{2} \text{ or } -6$$

2. Solve $9x^2 - 10x + 5 = 0$.

$$9x^2 - 10x - 5 = 0$$

$$(3x - 5)(3x + 1) = 0$$

$$x = \frac{5}{3} \text{ or } -\frac{1}{3}$$

3. A box has a base that measures 3 inches by 5 inches, and a height of 11 inches. Increasing each dimension of the base by the same amount causes the volume of the box to increase by 528 cubic inches.

- a. Using x as the amount that each dimension of the base is increased by, write an expression for the volume of the box. Let x be the increase in each dimension of the base in inches.

$$11(3+x)(5+x)$$

- b. Use the original measurements and the expression you wrote in part (a) to write an equation that can be used to find the amount by which each dimension of the base increased.

$$(3)(5)(11) + 528 = 11(3+x)(5+x)$$

- c. Solve the equation in part (b) and interpret the solutions in the context of the problem.

$$165 + 528 = 11(3+x)(5+x)$$

$$693 = 11(3+x)(5+x)$$

$$63 = (3+x)(5+x)$$

$$15 + 8x + x^2 = 63$$

$$x^2 + 8x - 48 = 0$$

$$(x+12)(x-4) = 0$$

$$x = -12 \text{ or } 4$$

Each dimension of the base is increased by 4 in.

Part B: Complete each problem.

4. Consider $9x^2 - 12x + 9 = -50$.

- a. Determine how many and what type of solutions the equation has.

$$b^2 - 4ac = (-12)^2 - 4(9)(9) = 144 - (36)(9) < 0$$

2 imaginary solutions

- b. Find those solutions by completing the square.

$$9x^2 - 12x + 9 = -50$$

$$9x^2 - 12x = -59$$

$$x^2 - \frac{4}{3}x = -\frac{59}{9}$$

$$x^2 - \frac{4}{3}x + \frac{16}{9} = -\frac{59}{9} + \frac{16}{9}$$

$$x^2 - \frac{4}{3}x + \frac{16}{9} = -\frac{43}{9} + \frac{16}{9}$$

$$(x - \frac{2}{3})^2 = -\frac{27}{9}$$

$$x - \frac{2}{3} = \pm \frac{\sqrt{27}}{3}$$

$$x = \frac{2}{3} \pm \frac{\sqrt{27}}{3}$$

5. A rocket is shot into the air from a 168-foot-high platform with an initial velocity of 68 feet per second.

- a. Use the quadratic formula to determine how much time the rocket is in the air before landing at ground level.

$$h(t) = -16t^2 + 68t + 168$$

$$t = \frac{-68 \pm \sqrt{68^2 - 4(-16)(168)}}{2(-16)}$$

$$= -1.5 \text{ or } 7$$

the rocket landed after 7 seconds

- b. Could the factoring method have been used to answer the question above? Explain.

Yes, because solutions are rational

Algebra 1 Cumulative Test 7a Answers

Karin Nielsen-Saines



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web molecular compounds many compounds do not contain ions but instead consist solely of discrete neutral molecules these molecular compounds covalent compounds result when atoms share rather than transfer gain or lose electrons covalent bonding is an important and extensive concept in chemistry and it will be treated in considerable

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web may 28 2020 answer problem 4 3 1 3 4 3 1 3 for each of the following compounds state whether it is ionic or covalent if it is ionic write the symbols for the ions involved a NF_3 b BaO c NH_4Cl d CO_3^{2-} e SrH_2 f PO_4^{3-} g IBr h Na_2O answer a answer b answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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web iron is an element it is made of only one type of atom iron atoms sulfur is another element it contains only sulfur atoms and nothing else so it is pure when we mix two different pure

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web jul 25 2022 write the name for each molecular compound CO_2 H_2S SF_6 N_2O strategy first you have to know how to name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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web molecular compounds are inorganic compounds that take the form of discrete molecules examples include such familiar substances as water H_2O H_2O and carbon dioxide CO_2 CO_2 these compounds are very different from ionic compounds like sodium chloride NaCl NaCl

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web may 28 2020 what is the total mass of hydrogen in each of the molecules a CH_4 b CHCl_3 c $\text{C}_{12}\text{H}_{10}\text{O}_6$ d $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b H_2O c CaNO_3 2 d $\text{CH}_3\text{CO}_2\text{H}$ acetic acid