

EE211 2023 Spring

Homework - Problem Set #2 - Solution

Problem 1. De Broglie wavelength (10 points)

Consider an electron has a kinetic energy of 0.1 eV .

a) Determine the wavelength of the electron.

(Sol)

The kinetic energy of free energy would be defined as:

$$E_{\text{kin}} = \frac{1}{2} m_e v^2 = 0.1\text{ eV}$$

Therefore, we can obtain the momentum of electron by

$$p = m_e v = \sqrt{2 m_e E_{\text{kin}}} = \sqrt{2 (9.109 \times 10^{-31} \text{ kg}) (0.1 \text{ eV})} = \sqrt{2 (9.109 \times 10^{-31} \text{ kg}) (0.1 \times 1.602 \times 10^{-19} \text{ J})} = \sqrt{2 (9.109 \times 10^{-31} \times 1.602 \times 10^{-20})}$$

Using De Broglie wavelength formula, we can obtain the wavelength of the electron by

$$\lambda = \frac{h}{p} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{1.708 \times 10^{-25} \text{ kg} \cdot \text{m} \cdot \text{s}^{-1}} = \frac{6.626 \times 10^{-34}}{1.708 \times 10^{-25}} \text{ m} = 3.879 \times 10^{-9} \text{ m} = 3.879 \text{ nm}$$

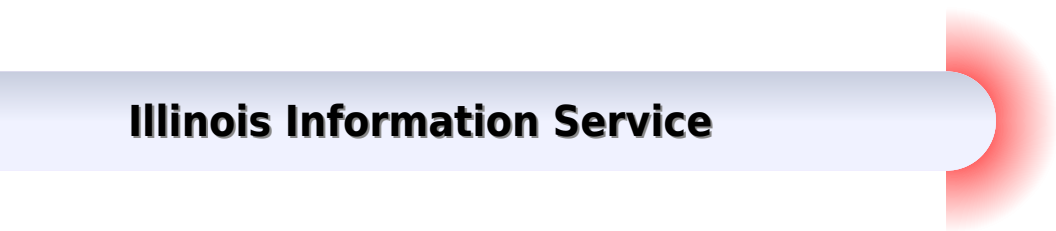
b) Explain why they are using an electron microscope to have higher magnification than optical one in terms of wavelength.

(Sol)

We can accelerate electrons to increase their energy. Then their wavelength become shorter than that of visible light photons (which range from 400 to 700nm). It allows electron microscopes to achieve higher magnification power than light microscopes.

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