

- (a) A particle P moves in a straight line with velocity $v \text{ ms}^{-1}$. At time $t = 0$, P is at the point O and has velocity 12 ms^{-1} . Its acceleration at time t seconds is given by $\frac{dv}{dt} = 3 \cos \frac{t}{4} \text{ ms}^{-2}$, ($t \geq 0$).
- Find an expression for the particle's velocity v , in terms of t .
 - Sketch a velocity/time graph for the particle for $0 \leq t \leq 8\pi$, showing clearly where the curve meets the axes and any maximum or minimum points.
 - Find the distance travelled by the particle before first coming to rest. [8]
- (b) Another particle Q moves in a straight line with displacement s metres and velocity $v \text{ ms}^{-1}$. Its acceleration is given by $a = -(v^2 + 4) \text{ ms}^{-2}$, ($0 \leq t \leq 1$). At time $t = 0$, Q is at the point O and has velocity 2 ms^{-1} .
- Show that the velocity v at time t is given by $v = 2 \tan\left(\frac{\pi - 8t}{4}\right)$.
 - Show that $\frac{dv}{ds} = -\frac{(v^2 + 4)}{v}$.
 - Find the distance travelled by the particle before coming to rest. [12]

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Simon Tennent



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$$\text{quad} \text{ Cu} + 2 \text{ AgNO}_3 \rightarrow 2 \text{ Ag} + \text{Cu}(\text{NO}_3)_2$$

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web a write a balanced equation for the reaction $\text{Al}_2\text{O}_3 + 4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$ b determine the theoretical yield of Al_2O_3 160.0 g Al $\times \frac{1 \text{ mol Al}}{27 \text{ g Al}} \times 2 \text{ mol Al}_2\text{O}_3 \times 101.964 \text{ g Al}_2\text{O}_3 = 302.3 \text{ g Al}_2\text{O}_3$ c determine the percent yield $\frac{\text{actual}}{\text{theoretical}} \times 100 = \frac{260.0 \text{ g}}{302.3 \text{ g}} \times 100 = 86.01\%$

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80 g I_2O_5 1 mol I_2O_5 1 mol I_2 xs 1 333.8 g I_2O_5 1 mol I_2O_5 28 g CO 1 mol CO

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