

The Student Placement Application Questionnaire

General Information

Applicant: _____
 Department: _____
 Address: _____
 Telephone: _____
 E-mail: _____

Student: _____
 Student ID: _____
 Program: _____
 Campus: _____
 Province: _____

General Training

English	French (Canada) / Canadian French	International English (British)	English as a Second Language (ESL) / French as a Second Language (FSL)
Supplementary	French for International English	French for International English (International English)	ESL / FSL
Additional courses in	French (Canada) / French (Canada)	English	ESL / FSL
Supplementary	Supplementary	Supplementary	
Additional courses	Supplementary	Supplementary	
Additional courses	Supplementary	Supplementary	

Additional

Additional courses in	Additional courses in English	Additional courses in English	Additional courses in English
Additional courses	Additional courses	Additional courses	Additional courses

Specific Skills

Additional English	Additional English	Additional English	Additional English
Additional English	Additional English	Additional English	Additional English
Additional English	Additional English	Additional English	Additional English
Additional English	Additional English	Additional English	Additional English
Additional English	Additional English	Additional English	Additional English
Additional English	Additional English	Additional English	Additional English

Additional / Other Remarks

Additional	Additional	Additional	Additional	Additional
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besuchen er findet dalton city vor welches gerade von den daltons mit dem saloon wieder in betrieb genommen wird mit
averell als koch dessen mahlzeiten jeder ausser rantanplan ungeniessbar findet

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web qualitative test flame tests introduction for this test i had five containers labelled a e each containing a metal salt i had
to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and
testing it by putting the metal under a roaring bunsen flame safety points wear googols all the time when

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web oct 15 2022 lab report flaming testing about exhibit is in the that paint watch in one flame tests are past to the
metallurgy and not the nonmetals in this combinations approved which heavy metal had monitored to expend

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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the unknown slat burned orange but also had a slight blue color close to the stick

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web oct 15 2022 the objectives of this lab are to perform light checks of metal cations in order to observe their typisch colors apply graphics to determine the periodicity and energizer of the emissions 5 flame tests and atomic spectra experiment chemistry libretexts flame test formal lab report sc4 ldc l1 how can atomic

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions lino3 cu no3 2 sr no3 2 ba no3 2 kno3 and nano3 since nitrates do not emit color you

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web task use a flame test to determine which ion salt produces the red color then using a spectroscope match the bright line spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material list causes red fireworks procedure

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web oct 15 2022 the objectives of this lab are to perform flame tests of metal cations in order to observe their characteristic colors perform calculations for determines the output and energy of the emitted

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web energy and matter answered worksheet the experiment conducted aimed to give students the opportunity to determine different metal ions in liquid and solid compounds by observing the color of the flames they give off when

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web sep 22 2021 pre laboratory assignment flame tests of metal cations in this lab you will perform flame tests of several different metal cations the characteristic colors observed are due to emitted electromagnetic radiation from the excited metal cations in this lab how do the metal cations become excited

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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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web 25 october 2022 objective experiment 1 to show how flame test can be used to identify some metals ion 2 to identify some unknown metal ions using flame test introduction the arrangement of electrons around the nucleus of the atom is

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