

Object 16 : Determine the boiling points of ethylene glycol - water solution of different compositions.

Introduction : In case of liquid mixtures, the boiling points depend upon their composition.

Consider a liquid mixture of ethylene glycol ($\text{CH}_2\text{OH} - \text{CH}_2\text{OH}$) and water (H_2O) of different composition by volume e.g. 10%, 20%, 30% of ethylene glycol by volume.

The boiling points are measured by taking 0.5-1 mL in a fusion glass tube and using a thermometer and Thiele's tube.

Object 17 : Determine the boiling

Determine The Boiling Points Of Ethylene Glycol Water Solution Of Different Compositions

Neil D. Jespersen, Alison Hyslop



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Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

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Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16
Ebook Chemistry: The Molecular Nature of Matter and Change 21st European Symposium on Computer Aided Process Engineering, 2011-06-10. The European Symposium on Computer Aided Process Engineering ESCAPE series presents the latest innovations and achievements of leading professionals from the industrial and academic communities. The ESCAPE series serves as a forum for engineers, scientists, researchers, managers and students to present and discuss progress being made in the area of computer aided process engineering. CAPE European industries large and small are bringing innovations into our lives whether in the form of new technologies to address environmental problems, new products to make our homes more comfortable and energy efficient or new therapies to improve the health and well being of European citizens. Moreover the European Industry needs to undertake research and technological initiatives in response to humanity's Grand Challenges described in the declaration of Lund namely Global Warming, Tightening Supplies of Energy, Water and Food, Ageing Societies, Public Health, Pandemics and Security. Thus the Technical Theme of ESCAPE 21 will be Process Systems Approaches for Addressing Grand Challenges in Energy, Environment, Health, Bioprocessing, Nanotechnologies.

General Chemistry James E. Brady, Gerard E. Humiston, 1980
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10 in One Study Package for CBSE Chemistry Class 12 with Objective Questions & 3 Sample Papers 4th Edition Disha Experts, 2020-06-20
Selection and Use of Engine Coolants and Cooling System Chemicals, 1963
Corrosion of Magnesium Alloys G L Song, 2011-03-25. The use of magnesium alloys is increasing in a range of applications and their popularity is growing wherever lightweight materials are needed. This book provides a comprehensive account of the corrosion of magnesium alloys. It covers not only the corrosion performances and mechanisms of Mg alloys in conventional environments such as sodium chloride solutions but also looks at their corrosion behaviours in special media like engine coolants and simulated body fluids. Part one covers fundamentals such as the corrosion, electrochemistry, activity and passivity of magnesium and its alloys. Part two then considers the metallurgical effect

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Physical Chemistry for the Biosciences Raymond Chang, 2005-02-11 This book is ideal for use in a one semester introductory course in physical chemistry for students of life sciences The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details Subsequently only basic skills of differential and integral calculus are required for understanding the equations The end of chapter problems have both physiochemical and biological applications

Coke & Chemistry, U.S.S.R., 1965 *Engine Coolant Testing* W. H. Ailor, 1980

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