



Determination Of % Available Chlorine In Bleaching Powder

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200





To determine the available chlorine in the given sample of bleaching powder by the iodometric method.

Introduction

- · Bleaching powder, also known as chlorinated lime
- yellowish-white powder having a smell of chlorine and is readily soluble in water
- It is prepared by passing chlorine gas over slaked lime at a temperature of 35-45° and consists of a mixture
 of calcium hypochlorite Ca(OCl₂) and calcium chloride CaCl₃, in addition some amount of free slaked lime
 i.e. Ca(OH)₂ H₂O is generally present
- Ca(OCl₂) is responsible for the bleaching action of bleaching powder. On treatment with glacial acetic acid, it liberates chlorine gas (Cl₂) as per the following reaction.

$$Ca(OCI,)+2CH,COOH \rightarrow Ca(CH,COO), +H,O+CI,$$

 The amount of chlorine liberated by the action of an acid on bleaching powder (CaOCl₂) is termed as available chlorine. The chlorine content of bleaching powder varies from 35 - 40%.





- bleaching powder also have strong germicidal and disinfectant properties.
- It finds application as a disinfectant for drinking water or swimming pool water.

Determination Of Available Chlorine In Bleaching Solution

Michael Brown

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A Text-book of Quantitative Chemical Analysis by Gravimetric, Electrolytic, Volumetric and Gastrometric Methods John Charles Olsen, 1905 A Text-book of Quantitative Chemical Analysis by Gravimetric, Electrolytic, Volumetric and Gasometric Methods John Charles Olsen,1916 A Text-book of Quantitative Chemical Analysis by Gravimetric, Ectrolytie, Volumetric and Gasometric Methods John Charles Olsen,1915 Lunge and Keane's Technical Methods of Chemical Analysis. 2d Ed., Edited by Charles A. Keane ... and P.C.L. Thorne Georg Lunge, 1924 Lunge and An Introductory course of quantitative **Keane's Technical Methods of Chemical Analysis** Georg Lunge,1924 **chemical analysis** Henry Paul Talbot, 1921 **Introductory Course of Quantitative Chemical Analysis** Henry P. Talbot.2019 Pulp and Paper Magazine of Canada ,1922 General Methods used in Technical Analysis, Electrolytic Methods of Chemical Analysis, Physical Measurements employed in Technical Analysis, Technical Gas Analysis, Fuel Analysis, Sulphurous Acid, Sulphuric Acid, and Nitric Acid, Saltcake and Hydrochloric Acid, Sodium Carbonate, The Chlorine Industry, Cyanogen Compounds Georg Lunge, 1924 Standard Methods of Chemical Analysis Wilfred Welday Scott, 1925 An Introductory Course of Quantitative Chemical Analysis Henry Paul Talbot, 2022-08-01 Henry Paul Talbot's An Introductory Course of Quantitative Chemical Analysis serves as a foundational text for students and practitioners in the field of chemistry With meticulous attention to detail Talbot navigates the multifaceted world of quantitative analysis providing clear explanations and practical examples The book s structured approach rich with illustrations and problem sets reflects its purpose to make complex concepts accessible while emphasizing rigorous scientific inquiry Written during a period when chemistry was rapidly evolving Talbot's work captures the essence of early 20th century scientific pedagogy merging traditional methods with emerging techniques in analytical chemistry Talbot an esteemed chemist and educator dedicated his career to advancing chemical education and fostering a deeper understanding of analytical methods His extensive experience in the laboratory and academia informed his writing enabling him to synthesize theoretical principles with hands on application His commitment to education is evident in his deliberate choice of language and pedagogical techniques which aim to equip beginners with the essential skills for success in quantitative analysis Recommended for students educators and practicing chemists this book stands as an invaluable resource for anyone eager to master quantitative analysis Talbot's clear exposition and rigorous approach not only lay the groundwork for practical application but also inspire a lifelong appreciation for the art and science of chemistry Bibliography of Pulp and Paper Making ,1929 Chemical & Metallurgical Engineering ,1922 Standard Methods of Chemical Analysis ,1922 Manual of Quantitative Chemical Analysis Joseph Otto Frank, Earl Amos Clemans, 1914 A Text-book of Quantitative Chemical Analysis ... Frank Julian, 1902 Iournal of the American Water Works Association .1925

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