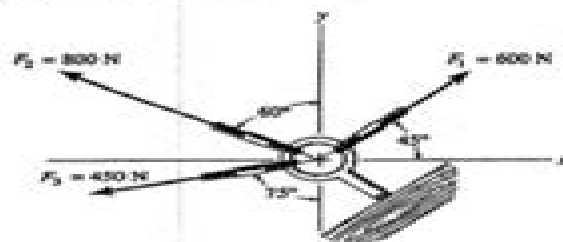


2-1. Determine the magnitude of the resultant force $F_R = F_1 + F_2 + F_3$ and its direction, measured counterclockwise from the positive x axis.



$$F_R = \sqrt{(800)^2 + (800)^2 - 2(800)(800)\cos 75^\circ} = 800.91 = 801 \text{ N} \quad \text{Ans}$$

$$\frac{800.91}{\sin 75^\circ} = \frac{800}{\sin \theta}$$

$$\theta = 63.05^\circ$$

$$\phi = 63.05^\circ + 45^\circ = 108^\circ \quad \text{Ans}$$



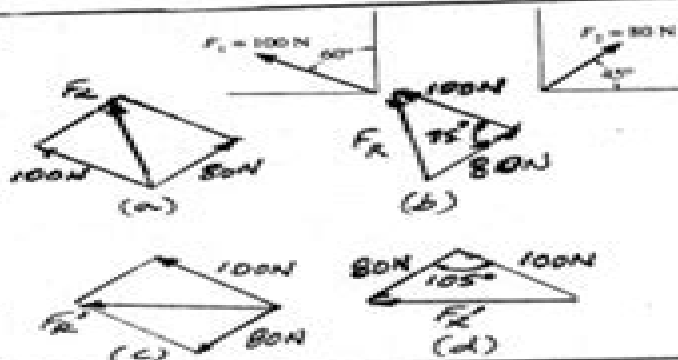
2-2. Determine the magnitude of the resultant force if: (a) $F_R = F_1 + F_2$; (b) $F_R = F_1 - F_2$.

Parallelogram Law: The parallelogram law of addition is shown in Fig. (a) and (c).

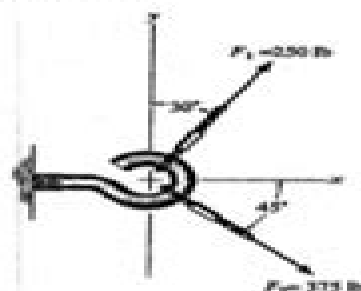
Trigonometry: Using law of cosines [Fig. (b) and (d)], we have

$$a) \quad F_R = \sqrt{100^2 + 80^2 - 2(100)(80)\cos 75^\circ} = 111 \text{ N} \quad \text{Ans}$$

$$b) \quad F_R' = \sqrt{100^2 + 80^2 - 2(100)(80)\cos 105^\circ} = 143 \text{ N} \quad \text{Ans}$$



2-3. Determine the magnitude of the resultant force $F_R = F_1 + F_2$ and its direction, measured counterclockwise from the positive x axis.



$$F_R = \sqrt{(250)^2 + (375)^2 - 2(250)(375)\cos 75^\circ} = 393.2 = 393 \text{ lb} \quad \text{Ans}$$

$$\frac{393.2}{\sin 75^\circ} = \frac{250}{\sin \theta}$$

$$\theta = 37.89^\circ$$

$$\phi = 360^\circ - 45^\circ + 37.89^\circ = 353^\circ \quad \text{Ans}$$



*2-4. Determine the magnitude of the resultant force $F_R = F_1 + F_2$ and its direction, measured clockwise from the positive x axis.



$$F_R = \sqrt{(500)^2 + (300)^2 - 2(500)(300)\cos 95^\circ} = 605.1 = 605 \text{ N} \quad \text{Ans}$$

$$\frac{605.1}{\sin 95^\circ} = \frac{300}{\sin \theta}$$

$$\theta = 33.40^\circ$$

$$\phi = 33.40^\circ + 30^\circ = 63.4^\circ \quad \text{Ans}$$



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kimia unsur golongan gas mulia dan halogen kimia - Apr 14 2022

golongan gas mulia dan halogen beberapa golongan unsur memiliki nama khusus seperti golongan viii a gas mulia dan viia halogen golongan tersebut memiliki nama khusus seperti sifat priodik sifat fisik dan sifat kimia kekhas sifat ke 2 golongan tersebut dapat membantu kita unsur unsur yang lain dalam 1 golongan semakin ke atas dan dalam

mengenal macam macam sifat dan unsur gas mulia kimia - Aug 31 2023

oct 7 2021 disebut gas mulia karena semua unsur pada golongan ini berwujud gas dan memiliki konfigurasi elektron yang sangat stabil sehingga akan sangat sulit untuk bereaksi dengan unsur lainnya unsur unsur gas mulia antara lain adalah helium he neon ne argon ar krypton kr xenon xe dan radon rn

sifat fisis dan kimia dari unsur gas mulia dan halogen ppt - May 16 2022

nov 26 2015 1 wujud gas mulia unsur gas mulia terdapat sebagai gas tak berwarna yang monoatomik ini erat kaitannya dengan struktur elektronoktet dan duplet dari gas mulia sedangkan wujud gas pada suhu kamar disebabkan titik

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nama helvy aprianty nim 291324951 fakultas tarbiyah dan keguruan pkm judul pengaruh model pembelajaran tgt terhadap hasil belajar siswa pada materi tata nama senyawa di mas darul ihsan aceh besar jadwal sidang rabu 05 juni 2017 jam 10 00 wib ruang 01 tebal skripsi 66 halaman pembimbing 1 ir amna emda m pd

gas mulia dan halogen by finlan aldan prezi - Jul 18 2022

oct 16 2014 sifat kimia dinamakan gas mulia karena kereaktifannya yang sangat rendah sempat disebut gas inert yang terbantahkan karena sudah ditemukan senyawa gas mulia dan gas langka yang terbantahkan karena argon ternyata mengisi sekitar 0 94 volume atmosfer bumi helium adalah gas mulia yang pertama ditemukan konfigurasi elektron

ppt gas mulia dan halogen kimia ppt slideshare - May 28 2023

apr 16 2017 mempunyai titik didih dan titik leleh yang sangat rendah jari jari atom gas mulia sangat kecil mempunyai elektron valensi 8 khusus untuk helium elektron valensinya 2 sudah stabil sehingga sukar membentuk senyawa dengan unsur lain dialam tidak pernah di temukan senyawa gas mulia gas mulia terdapat sebagai molekul monoatomik

makalah tentang gas mulia dan halogen pdf scribd - Apr 26 2023

fgas mulia dan halogen sejarah gas mulia golongan gas mulia atau golongan viii a adalah unsure unsur yang memiliki delapan elektron valensi dengan konfigurasi elektronik terluar $ns^2 np^6$ unsur unsur tersebut adalah helium he neon

gas mulia pengertian sifat kegunaan pembuatan konfigurasi - Jun 28 2023

oct 6 2023 gas mulia ditemukan di dekat golongan halogen dalam tabel periodik karena unsur gas mulia memiliki konfigurasi elektron yang penuh unsur unsur tersebut tidak reaktif dan senyawanya tidak dikenal akibatnya gas gas ini dikenal dengan gas inert

apa itu halogen dalam kimia berikut definisi unsur sifat dan - Nov 21 2022

may 18 2022 dikutip dari buku kimia unsur golongan gas mulia halogen alkali dan alkali tanah yang disusun oleh tantri ambarsari 2020 halogen adalah senyawa yang unsur unsurnya dapat bereaksi dengan logam untuk membentuk garam reaktivitas unsur halogen sangat besar sehingga unsur unsurnya sulit ditemukan di alam