

practice test ch 7 & 8

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following best describes the rates of chemical reaction?
- most chemical reactions occur very slowly
 - most chemical reactions occur at moderate rates
 - most chemical reactions occur very rapidly
 - chemical reactions have a wide range of rates, from extremely fast to extremely slow
- _____ 2. Which of the following is true of the rates of most chemical reactions?
- the initial rate is cannot be measured
 - the initial rate is faster than the rate later in time
 - the initial rate is the same as the rate later in time
 - the initial rate is slower than the rate later in time
- _____ 3. Which of the following is true of effective collisions?
- the number of effective collisions decreases as the temperature is increased
 - the number of effective collisions determines the reaction rate
 - both a. and b.
 - Neither a. nor b.
- _____ 4. Which of the following is true of effective collisions?
- the number of effective collisions determines the reaction rate
 - the number of effective collisions increases as the temperature is increased
 - both a. and b.
 - Neither a. nor b.
- _____ 5. Which of the following is true?
- decreasing the activation increases the reaction rate
 - the activation energy for an exothermic reaction is negative
 - both a. and b.
 - Neither a. nor b.
- _____ 6. In an energy diagram for a chemical reaction what species appears at the highest energy?
- the catalyst
 - the products
 - the reactants
 - the transition state
- _____ 7. Given that the reaction $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$ is exothermic, which of the following is true of the reaction $2 \text{H}_2\text{O}(\text{g}) \rightarrow 2 \text{H}_2(\text{g}) + \text{O}_2(\text{g})$?
- its activation energy is lower than that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - its activation energy is higher than that of c. its activation energy is the same as that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
 - there is no relationship between its activation energy and that of $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{g})$
- _____ 8. In an energy diagram for an endothermic chemical reaction which of the following is true?
- the energy of the products is lower than that of the reactants
 - the energy of the transition state is higher than that of the reactants
 - both a. and b.
 - Neither a. nor b.
- _____ 9. Consider a reaction such as $\text{A}_2(\text{g}) + \text{B}_2(\text{g}) \rightarrow 2\text{AB}(\text{g})$. What can we say about the mechanism of the reaction?
- if the reaction is fast it occurs in a single step
 - if the reaction is fast it occurs in multiple steps
 - if the reaction is slow it occurs in multiple steps

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Michael Hoelscher



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