	solution has more H* ion concentration ?		(B) 3,3-Dimethylbutan-2-ol
	(1) A		(C) 2,3,3-Trimethylbutan-2-ol
	(2) B		(1) They are allotropes.
	(3) Both have equal concentration		(2) They are isomers.
21.	(4) Cannot be determined.	 (3) They are homologoes. (4) They are isobars. 26. An element X has electronic configuration 2, 8, 3. What is true about this element? (A) Its atomic radius will be more than that of an element Y having electronic configuration 2, 8, 7. (B) The oxide formed by X will react with aqueous solution of hydroxide of an element with electronic configuration 2, 8, 1. (C) It lies in the same group as Boron (metalloid) and gallium. (1) A & B (2) B & C (3) Only B (4) A, B & C 27. Which of the following properties generally increases on moving from top to bottom in a group? (1) Ionization energy (2) Non-metallic character 	
	An aqueous solution turns red litmus solution blue. Excess addition of which of the following solutions would reverse the change? (1) Baking powder (2) Lime (3) Ammonium hydroxide sloution		An element X has electronic configuration 2, 8, 3. What is true about this element? (A) Its atomic radius will be more than that of an element Y having electronic configuration 2, 8, 7.
22.	(4) Hydrochloric acid When a mixture of iron oxide and aluminium powder, is heated, intense heat is generated. This fact is made use of in (1) Soldering (2) Welding of steel (3) Making iron (4) Cutting of metals		
23.	Which of the following statements is/are correct about iron? (A)Stainless steel is an alloy of nickel, chromium and iron.		increases on moving from top to bottom in a group ? (1) Ionization energy

How are the following related to each other ?

(A) 3-Methylbutan-2-ol

20. You have two solutions, A and B. The pH of 25.

solution A is 6 and pH of solution B is 8. Which

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