



## Mole Ratios

## Copper and Silver Nitrate

#### Introduction

The reaction of copper wire with office retrace in acuseus solution provides an interesting display of chemistry in action—celecate allow crystals begin to grow on the wire surface, and the color of copper(II) torus gradually appears in adaption. What relationships govern the relative cuantities of reactants and products in this chemical reaction?

## Concepts

Néferèn elation.

- Habitraria obsessional economics;
- Suoichiometric
- · Single replacement reactions

## Background

Stockhowerry is the branch of chemistry that disds with the numerical relationships and mathematical proportions of reactants and products in a chemical reaction. One of the most important become of statichiometry is that the amounts of the reactants and products in a chemical reaction are related to one another on a mode basis. Chemical reactions are normally represented by bulanced chemical equations. The coefficients in a balanced demaction against the relative number of modes of each reactant, and product involved in the reaction. The response of these coefficients represent the each toxic that govern the disappearance of reactants and appearance of products. Knowing the mode ratio in a balanced chemical equation is governed to solved to showing the reader attoring a balanced chemical equation is governed to solved to solved products y problems.

The reaction of supper metal with eller nitrate solution is a single replacement reaction, represented by the following unbalanced chemical exaction.

$$a \operatorname{Cutal} + A_0 \operatorname{NO}_0(ac) \rightarrow c \operatorname{CutNO}_0(ac) + A_0 \operatorname{Cut}$$

The values of the coefficients  $x, \theta, c$  and x can be determined experimentally by measuring the mass of copper wire that reacts and the mass of metallic eliver that is produced in the reaction.

#### Experiment Overview

The purpose of this experiment is to determine the number of moles of tractants and products in the reaction of capper and effect ritrate and to calculate their node ratio. The node ratio relating the chapper and of capper and the formation of effect well be used to write the half-artest chemical equation for the resistion.

## Pre-Lab Questions

Copper(II) chloride (CuCl<sub>2</sub>: 0.98 g) was dissolved in mater, and a piece of aluminum wire (Al; 0.56 g) was placed in the solution. The blue color due to copper(II) chloride score faire, and a red precipitate of solid copper was observed. After the blue color had disappeared completely, the leftwore aluminum wire was removed; from the solution and weighted. The mass of the leftwore aluminum wire was removed; from the solution and weighted.

- Calculate the number of nodes of copper(II) chloride and of duminum this resided.
- What is the moder axio of copper(II) chloride to aluminum metal?
  Extress this to the neurost whole manifest ratio.
- What happened to the abordinary metal that was consumed in this reaction? Write the formula of the next probable aluminary constining product.
- Write a halanced chemical equation for the single replacement rescrition of apper (II) oblimite with abunings.

#### Materials

Acquine, 125-rull, in a 250-rull, beaker\*

Copper wire, Ca. 25 and

Nitric acid, HNO, 3 M, 3 drops.

Silver mitrate, AgNO, 1.5 g.

Hallance, contiguant (0.01-g; precision).

Heakers, 50- and, 100-oid,

Erlermeyer (look, 125-ml.

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**Kathleen Armour** 

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "Silver Nitrate Lab Report Mole Ratio Answers Wangpoore," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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