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students did not possess Also because MATLAB was originally adopted by practitioners in the fields of signal processing and electrical engineering most of these texts provided examples primarily from those areas an approach that didn't fit with a general engineering curriculum This text starts with basic algebra and shows how MATLAB can be used to solve engineering problems from a wide range of disciplines The examples are drawn from concepts introduced in early chemistry and physics classes and freshman and sophomore engineering classes A standard problem solving methodology is used consistently The text assumes that the student has a basic understanding of college algebra and has been introduced to trigonometric concepts students who are mathematically more advanced generally progress through the material more rapidly Although the text is not intended to teach subjects such as statistics or matrix algebra when the MATLAB techniques related to these subjects are introduced a brief background is included In addition sections describing MATLAB techniques for solving problems by means of calculus and differential equations are introduced near the end of appropriate chapters These sections can be assigned for additional study to students with a more advanced mathematics background or they may be useful as reference material as students progress through an engineering curriculum

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alcohols ethanol 2 methylpropan 2 ol butan 2 ol but 2 en 1 ol ch₃ch₂ch₂ch₂oh ii ch₃ ch₃ch₂ch₂oh iii ch₃

ch₃ch₂ch₂oh iv ch₃ ch₃cch₂oh ch₃ 6 aldehydes and ketones draw the structures for the following compounds ethanal butanone also known as butan 2 one but the 2 is unnecessary and is often left out 2 methylpropanal

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons
heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with
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