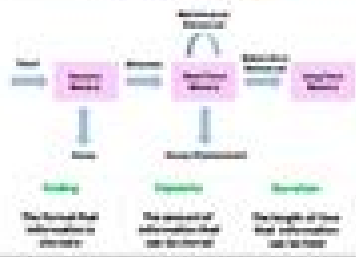


PSYCHOLOGY PAPER 1:

MEMORY

Multi-Store Model



MEMORY

Working Memory Model



PSYCHOPATHOLOGY

Definition of Abnormality

A person may be judged to be abnormal if they have a condition that causes them to behave in a way that is different from the majority of the population.

Statistical Infrequency
A person may be judged to be abnormal if they have a condition that causes them to behave in a way that is different from the majority of the population.

Deviation from Social Norms
A person may be judged to be abnormal if they have a condition that causes them to behave in a way that is different from the majority of the population.

Personal Distress
A person may be judged to be abnormal if they have a condition that causes them to behave in a way that is different from the majority of the population.

PSYCHOPATHOLOGY

Characteristics

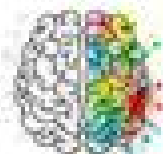


Characteristics of OCD

OCD is a mental health condition that causes a person to have obsessive thoughts and compulsive behaviors.

AQA PSYCHOLOGY PAPER 1

Revision Notes



SOCIAL INFLUENCE

Conformity



Definition
Conformity is the change in a person's behavior or beliefs as a result of real or imagined social pressure.

Asch's Line Judgment Task
Asch's line judgment task is a classic experiment in social psychology. It involves a group of people being asked to judge the length of a line. One person is the real participant, and the others are confederates. The confederates are instructed to give a wrong answer, and the real participant is asked to give their answer. The results show that the real participant often conforms to the wrong answer given by the confederates.

Factors Affecting Conformity
There are several factors that can affect conformity. These include the size of the group, the unanimity of the group, the status of the individual, and the task itself.

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Nicholas Alexandros Savva



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web in today s experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightarrow \text{Ca}^{2+} + 2\text{OH}^-$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ 0 M hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ the solubility product was determined to be ksp $\text{Ca}^{2+} + 2\text{OH}^-$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3(\text{PO}_4)_2$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $\text{Ca}_3(\text{PO}_4)_2 \rightleftharpoons 3\text{Ca}^{2+} + 2\text{PO}_4^{3-}$ $K_{\text{sp}} \text{Ca}_3(\text{PO}_4)_2$

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated Ca(OH)_2 soultion with a st andard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web question determining the ksp of calciumhydroxide Ca(OH)_2 s Ca^{2+} aq 2OH^- aq for this experiment the titration of calciumhydroxide with hydrochloric acid was investigated determining the ksp of calcium hydroxide Ca(OH)_2 s Ca^{2+} aq 2OH^- aq for this experiment the titration of calcium

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for Ca(OH)_2 was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide Ca(OH)_2 s Ca^{2+} aq 2OH^- aq $K_{sp} = 5 \times 10^{-6}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for ksp for sodium hydroxide is 5×10^{-6}

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