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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below

$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web you may have to consult the solubility rules write the net ionic equation for the reaction of $\text{BaCl}_2(\text{aq})$ and $\text{Na}_2\text{SO}_4(\text{aq})$ you may have to consult the solubility rules write

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a $\text{K}_2\text{CO}_3(\text{aq}) + 2\text{CuCl}(\text{aq})$

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web the net ionic equation for equation 1 is given as equation 3 s2 aq fe2 aq gt fes s eq 3 8 note that the phase indicator aq is optional on ionic species but is

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