

KEY

Chemistry Worksheet: Matter #1

1. A mixture (~~is~~ is not) a chemical combining of substances.
2. In a compound the (atoms ~~molecules~~) are (chemically ~~physically~~) combined so that the elements that make up the compound (retain ~~lose~~) their identities and (~~do~~ do not) take on a new set of properties.
3. The smallest identifiable unit of a compound is a(n) molecule, which is made up of atoms _{elements} which are chemically bonded.
4. True or False: A mixture is always made up of a combination of elements. *or compounds*
5. In a mixture, the substances (~~lose~~ retain) their identities.
6. In a mixture the substances involved (can ~~cannot~~) be separated by a simple physical process.
In a compound the elements involved (can ~~cannot~~) be separated by a simple physical process because the elements are (physically combined ~~chemically bonded~~).
7. True or False: An element can be broken down into a simpler substance.
8. The smallest identifiable unit of an element is a(n) atom.
9. From the following list of substances, circle the ones that are elements:

<u>silver</u>	carbon dioxide	wood alcohol	<u>chromium</u>
<u>water</u>	<u>hydrogen</u>	<u>carbon</u>	<u>nitrogen</u>
<u>oxygen</u>	<u>gold</u>	sugar	salt
air	<u>sulfur</u>	<u>magnesium</u>	<u>nickel</u>
10. Explain how to separate the sugar and water in a solution of sugar and water.
Boil the water to leave the sugar behind.
11. How would you separate a mixture of alcohol and water?
Freeze the mixture, water will freeze and alcohol will remain a liquid.
12. How would you separate sand and water?

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Chris Sunley



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Chemistry, Study Guide James E. Brady, John R. Holum, 1993-02-11 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples

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Prentice Hall Science Explorer Michael J. Padilla, 2002 Cambridge IGCSE™ Chemistry Teacher's Guide (Collins Cambridge IGCSE™) Chris Sunley, 2022-02-03 Prepare students with complete coverage of the revised Cambridge IGCSE™ Chemistry syllabus 0620 0971 for examination from 2023 Collins Cambridge IGCSE Chemistry Teacher's Guide is full of lesson ideas practical instructions technician's notes planning support and more

Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Katherine, 2006
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Study Guide [to Accompany] General Chemistry Dixie Goss, 2002 **Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

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