

Part I - Listening

Directions for items 1-25. You will hear questions on the test tape. Select the one correct answer. (ONE answer only.)

1. _____ .

- ☐ twenty-four
- ☐ lemons
- ☐ twelve
- ☐ six

2. _____ .

- ☐ ice
- ☐ fog
- ☐ rain
- ☐ humidity

3. _____ .

- ☐ the outside
- ☐ the frame
- ☐ The upstairs
- ☐ the inside

4. _____ .

- ☐ changes
- ☐ buttons
- ☐ pockets
- ☐ trousers

5. _____ .

- ☐ his fever
- ☐ his throat
- ☐ his ears
- ☐ his heart beat

6. _____ .

- ☐ yesterday
- ☐ a week ago
- ☐ two days ago
- ☐ three days ago

7. _____ .

- ☐ rainy
- ☐ dark
- ☐ cold
- ☐ clear

8. _____ .

- ☐ making surfaces smoother
- ☐ sawing wood
- ☐ cutting metal
- ☐ making holes

9. _____ .

- ☐ next week
- ☐ after a few minutes
- ☐ some time later
- ☐ immediately

10. _____ .

- ☐ by check
- ☐ in cash
- ☐ before I stayed in the room.
- ☐ after I stayed in the room

11. _____ .

- ☐ It broke the window.
- ☐ It opened the window.
- ☐ It dented the window.
- ☐ It closed the window.

12. _____ .

- ☐ only \$5,000
- ☐ less than \$5,000
- ☐ \$5,000 or higher
- ☐ just \$5,000

Alcpt Test Example

Amanda Jefferies and Marija Cubric

A red circular graphic with a gradient, appearing as a partial circle or a thick arrow pointing to the right, located on the right side of the slide.

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Carbon-14 emits beta radiation and decays with a half-life ($t_{1/2}$) of 5730 years. Assume you start with a mass of 2.00 g of carbon-14. a. How long is ... ECON101 - Ch.25 Section Review Answers For the electronic transition from $n = 3$ to $n = 5$ in the hydrogen atom. a) Calculate the energy. b) Calculate the wavelength (in nm). Chapter 25 Nuclear Chemistry 25.2 Nuclear Transformations Sep 5, 2017 — Nuclear Chemistry Targets: 1. I CAN Utilize appropriate scientific vocabulary to explain scientific concepts. 2. I CAN Distinguish between fission ... Matter and Change • Chapter 25 When a radioactive nucleus gives off a gamma ray, its atomic number increases by 1. 2. The three types of radiation were first identified by Ernest Rutherford. Nuclear Chemistry - Lake Central High School Jul 12, 2015 — What is the change in atomic number after the alpha decay? It decreases by 2. b. ... answer the following questions. **Nuclear** ... 25.2 Nuclear Transformations | Lecture notes Chemistry These nuclei decay by turning a neutron into a proton to emit a beta particle (an electron) from the nucleus. This process is known as beta emission. It ... 60 s - 1 min SECTION 25.2 NUCLEAR TRANSFORMATIONS. 1. Write a nuclear equation for the following radioactive processes. a. alpha decay of francium-208 $^{208}\text{Fr} \rightarrow \text{b} \dots$