

Name _____ Hour _____

Practice 17: Classifying and Balancing Chemical Reactions

- Balance each reaction (IF THEY ARE NOT BALANCED ALREADY).
- Classify each reaction as decomposition, synthesis, single replacement, double replacement, or combustion.

BALANCE REACTION	CLASSIFY REACTION
1. $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$	_____
2. $\text{H}_2\text{SO}_4 + \text{Fe} \rightarrow \text{H}_2 + \text{FeSO}_4$	_____
3. $\text{Zn} + \text{S} \rightarrow \text{ZnS}$	_____
4. $\text{Al}_2\text{S}_3 \rightarrow \text{Al} + \text{S}$	_____
5. $\text{C}_{12}\text{H}_{22}\text{O}_{11} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$	_____
6. $\text{NaOH} + \text{CuSO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{Cu(OH)}_2$	_____
7. $\text{Cl}_2 + \text{KBr} \rightarrow \text{KCl} + \text{Br}_2$	_____
8. $\text{C}_4\text{H}_{12} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$	_____
9. $\text{Mg}_3(\text{PO}_4)_2 + \text{H}_2 \rightarrow \text{Mg} + \text{H}_3\text{PO}_4$	_____
10. $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \text{H}_2\text{O}$	_____

Classifying Chemical Reactions Answers Physical Science If8767

N Colangelo



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Exploring Chemical Reactions Nigel Saunders,2007 *The Basics of Chemical Reactions* Krista West,2013-12-15
Explores the simplicity of basic chemical reactions and then builds to the more complex giving readers a history of the years and the minds that contributed to the research that led to chemistry as we know it today Atoms and Chemical Reactions
Suzanne Slade,2007-08-01 Examines chemical reactions chemical equations atoms and molecules *Chemical Reactions and Their Equations* Ingo Waldemar Dagobert Hackh,1921 Physical and Chemical Changes (ENHANCED eBook) Edward P. Ortleb,Richard Cadice,1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter The definition of matter is presented along with explanations of states and properties of matter Topics include atoms molecules elements compounds mixtures solutions symbols and formulas Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key **Real World Examples of Four Types of Chemical Reactions** , James W Smith developed Real World Examples of Four Types of Chemical Reactions a physical science and chemistry lesson for 9th grade students The students use the Internet to locate information about industrial pharmaceutical and environmental operations that use examples of the four general classes of chemical reactions These reactions are synthesis decomposition single replacement reactions and double replacement reactions The Louisiana Challenge Grant provides the lesson online as part of the Louisiana Challenge collection of activities for the K 12 classroom **Chemical Reactions** Denise Walker,2007 This title introduces the reader to the huge variety of chemical reactions that shape our world Find out all about explosions learn about how to start reactions and understand how chemical equations work *Chemical Reactions* John A. Olson,2011-06-08
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Understanding Chemistry Robert J. Ouellette,1987 **Chemicals & Reactions** Jon Richards,2004-08 Chemical reactions are happening around us and inside us every day Every time we boil water wash our hands or stir sugar into cocoa a chemical reaction is taking place Book jacket **Reactions** Theodore W. Gray,2017 The book begins with a brief recap of

elements and molecules and then goes on to explain important concepts that characterize a chemical reaction including Energy Entropy and Time. It is then organized by type of reaction including Combustion Photosynthesis Respiration Oxidation and Fermentation. A special section dedicated to chemical cycles includes The Carbon Cycle The Iron Cycle and The Lime Cycle. *Chemical Reactions* Eve Hartman, Wendy Meshbesher, 2009-07-10 *Chemical Reactions* Nigel Saunders, 2007. This book explains what chemical reactions are what elements atoms and molecules are and what happens to them during chemical reactions. **Chemical Reactions** Louise Spilsbury, Richard Spilsbury, 2007. Readers will learn what chemical reactions are how they work what changes happen during reactions and how we can stop reactions. *Encyclopedia of Chemical Reactions* Carl Alfred Jacobson, 1946.

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