

The appropriate K_p or K_c values can be found in your textbook.

Q10) What are the conjugate bases of the following acids?

a) HNO_2 NO_2^-

b) HNO_3 NO_3^-

c) $H_2PO_4^-$ HPO_4^{2-}

d) $H_2C_2O_4$ $C_2O_4^{2-}$

Q11) What are the conjugate acids for each of the bases?

a) OH^- H_2O

b) $C_2O_4^{2-}$ $H_2C_2O_4$

c) ClO_3^- $HClO_3$

d) NO_2^- HNO_2

Q12) Identify the Brønsted-Lowry acids and the Brønsted-Lowry bases. The conjugate base and the conjugate acid of the following molecules:

a) HNO_2 + OH^- \rightleftharpoons NO_2^- + H_2O

acid base base acid

b) HNO_2 + H_2O \rightleftharpoons NO_2^- + H_3O^+

acid base base acid

c) NO_2^- + H_2O \rightleftharpoons HNO_2 + OH^-

base acid acid base

d) H^+ + OH^- \rightleftharpoons H_2O

acid base acid

e) H_2CO_3 + PO_4^{3-} \rightleftharpoons HCO_3^- + HPO_4^{2-}

acid base base acid

f) H_2O + H_2CO_3 \rightleftharpoons H_3O^+ + HCO_3^-

base acid acid base

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