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web balance each of the following reactions and identify each type of reaction $\text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + \text{NaOH}$ $\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{NH}_4^+ + \text{SO}_4^{2-}$
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web 5 in each of the reactions in model 1 there are ions present in the solutions that do not participate in the chemical reaction in other words they exist in the spectator form both

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web identify the type of chemical reaction from model 1 that would describe this reaction b write chemical formulas in the box provided below of the substances in the word

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