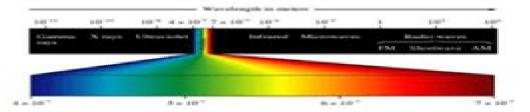
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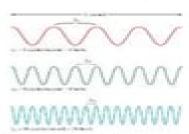
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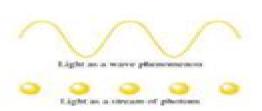
I. a) What is the electromagnetic spectrum?



b) What distinguishes one type of EMR from another?

→ Wavelength and frequency are ______ related – as the frequency of the EMR increases, the wavelength and vice versa.





2. How is the energy of a photon calculated?

Example:

Ultraviolet (UV) light that causes tanning and burning of the skin has a higher energy per photon than infrared (IR) light from a heat lamp.

*Calculate the energy of a 1.5 x 10¹³ Hz *UV photon* and a 3.3. x 10¹⁴ Hz *IR photon*. → energy of UV photon: → energy of IR photon:

*Therefore, the energy of a photon is solely dependant on the ______ of the EMR.

(i.e. higher frequency = more _____)

(e.g. E of UV photon > E of blue light photon > E of red light photon > E of infrared photon etc.)

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