

Chemistry 105

Chapter 3 Review Problems

- How many molecules are present in 180 grams of water? (The atomic weights of H and O are 1 g/mol and 16 g/mol, respectively.)
a) 6.0×10^{24}
b) 10
c) 5.6×10^{-28}
d) 1.8×10^2
e) 180×10^{28}
- The atomic mass of sulfur is given as 32.07 g/mol. What is the mass of an average atom of sulfur?
a) 5.33×10^{-23} g
b) 32.07 g
c) 32.07×10^{-23} kg
d) 193×10^{-23} g
e) 6.02×10^{23} g
- What is the mass of C present in 60 g of C_3H_8 ?
a) 12 g
b) 24 g
c) 48 g
d) 60 g
e) 6 g
- What is the molar mass of C_3H_8O ?
a) 24 g/mol
b) 32 g/mol
c) 7 g/mol
d) 35 g/mol
e) 46 g/mol
- What is the correct percentage by mass of nitrogen in HCN?
a) 33 %
b) 14%
c) 27%
d) 52%
e) 98%
- What is the chief danger that is anticipated if CO_2 levels get too high in the atmosphere?
a) people will develop skin cancer from exposure to UV light
b) the atmosphere will develop an unpleasant odor
c) the temperature of the planet may climb by several degrees
d) plant life will be exterminated
e) The CO_2 will remove ozone from the stratosphere

Chapter 3 Chemistry Test Answers

Thomas Evangelist



Chapter 3 Chemistry Test Answers:

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Class 9 Chemistry MCQ (Multiple Choice Questions)

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Reviews topics covered on the test offers tips on test taking strategies and includes two full length practice tests with answers and explanations *McGraw-Hill's 500 MCAT General Chemistry Questions to Know by Test Day* John T.

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