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to accompany

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THIRD EDITION



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that occur in practice Therefore basic concepts with examples and many exercises are presented in each topic instead of specific projects of the industry The main objective was to provoke students to observe kinetic phenomena and to think about them Indeed reactors cannot be designed and operated without knowledge of kinetics Additionally the empirical nature of kinetic studies is recognized in the present edition of the book For this reason analyses related to how experimental errors affect kinetic studies are performed and illustrated with actual data Particularly analytical and numerical solutions are derived to represent the uncertainties of reactant conversions in distinct scenarios and are used to analyze the quality of the obtained parameter estimates Consequently new topics that focus on the development of analytical and numerical procedures for more accurate description of experimental errors in reaction systems and of estimates of kinetic parameters have been included in this version of the book Finally kinetics requires knowledge that must be complemented and tested in the laboratory Therefore practical examples of reactions performed in bench and semi pilot scales are discussed in the final chapter This edition of the book has been organized in two parts In the first part a thorough discussion regarding reaction kinetics is presented In the second part basic equations are derived and used to represent the performances of batch and continuous ideal reactors isothermal and non isothermal reaction systems and homogeneous and heterogeneous reactor vessels as illustrated with several examples and exercises This textbook will be of great value to undergraduate and graduate students in chemical engineering as well as to graduate students in and researchers of kinetics and catalysis

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