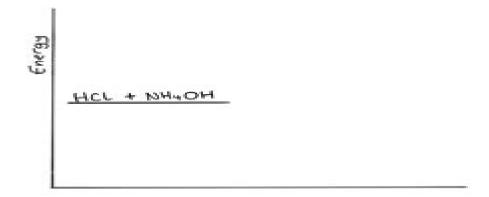
7. In an exothermic reaction, is enthalpy change positive or negative?

8. In an endothermic reaction, is enthalpy change positive or negative?

When hydrochloric acid reacts with ammonium hydroxide in a beaker, the temperature goes up.

$$\Delta H = -53.4 \text{kJ/mol}$$

Complete the energy profile diagram and state whether the reaction is endothermic or exothermic, explain your answer.



10. What are the units for enthalpy change, ΔH

You may like to watch this lesson on the different types of Chemical reactions.



Energy Changes in Chemical reactions for KS3 Science

Chemical Reactions And Energy Worksheet Answers

Albert Kaskel

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