

16. Given the chemical formula of an ionic compound, list at least three necessary steps to give the correct name of that compound. (If needed, use a chemical formula of a compound from the table above as an example in listing the naming steps.)

CHEMICAL BONDING POGIL

Activity 4 – Predicting the correct chemical formula for ionic compounds formed from simple anions

Objective:

- To learn how to predict the correct number of cations and anions in a simple salt.

The Model:

Table 3

Cations	Anions	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Na^+	S^{2-}	Na_2S	sodium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Exploring the Model

1. What is the charge on the zinc ion?
2. What is the charge on the nitride ion?
3. What is the charge on the chloride ion?
4. What is the charge on the ionic compound, sodium chloride?
5. What is the charge on the ionic compound, sodium sulfide?
6. How many potassium ions are present in K_3N ?
7. What does the “2” stand for in the formula for ZnCl_2 ?

Exploring the Model

8. Sodium chloride is NaCl , and zinc chloride is ZnCl_2 . Why are there more chloride ions in the zinc compound?
9. Sodium chloride is NaCl , and sodium sulfide is Na_2S . Why are there more sodium ions in the sulfide compound?

Chemical Bonding Pogil Answers Key

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1. Understanding the eBook Chemical Bonding Pogil Answers Key
 - The Rise of Digital Reading Chemical Bonding Pogil Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bonding Pogil Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Bonding Pogil Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bonding Pogil Answers Key
 - Personalized Recommendations
 - Chemical Bonding Pogil Answers Key User Reviews and Ratings
 - Chemical Bonding Pogil Answers Key and Bestseller Lists

5. Accessing Chemical Bonding Pogil Answers Key Free and Paid eBooks
 - Chemical Bonding Pogil Answers Key Public Domain eBooks
 - Chemical Bonding Pogil Answers Key eBook Subscription Services
 - Chemical Bonding Pogil Answers Key Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Chemical Bonding Pogil Answers Key Compatibility with Devices
 - Chemical Bonding Pogil Answers Key Enhanced eBook Features
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 - Highlighting and Note-Taking Chemical Bonding Pogil Answers Key
 - Interactive Elements Chemical Bonding Pogil Answers Key
8. Staying Engaged with Chemical Bonding Pogil Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Bonding Pogil Answers Key
9. Balancing eBooks and Physical Books Chemical Bonding Pogil Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Bonding Pogil Answers Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Bonding Pogil Answers Key
 - Setting Reading Goals Chemical Bonding Pogil Answers Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Bonding Pogil Answers Key
 - Fact-Checking eBook Content of Chemical Bonding Pogil Answers Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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