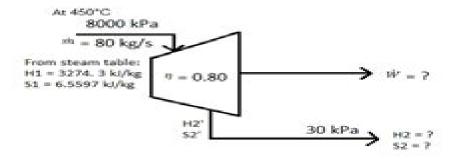
Introduction to Chemical Engineering Thermodynamics Problem Set

A turbine operates adiabatically with superheated steam entering at 450°C and 8,000 kPa with a mass flowrate of 80 kg/s. The exhaust pressure is 30 kPa and the turbine efficiency is 0.80. Determine the power output of the turbine and the enthalpy and entropy of the exhaust steam.

Given:



Required:

- a. W. power output of the turbine
- H₂, enthalpy of exhaust stream
- c. S2, entropy of exhaust stream

Solution:

From steam table under superheated steam, Table F.2, for inlet conditions:

$$H_1 = 3274.3 \frac{kJ}{kg}$$

 $S_1 = 6.5597 \frac{kJ}{kg \cdot K}$

For discharge conditions:

$$H_2^L = 289.302 \frac{kJ}{kg}, H_2^p = 2625.4 \frac{kJ}{kg}$$

 $S_2^L = 0.9441 \frac{kJ}{kg \cdot K}, S_2^p = 7.7695 \frac{kJ}{kg \cdot K}$

If the expansion at 30 kPa is isentropic, then: $S_2' = S_1 = 6.5597 \frac{kJ}{kg\cdot K}$. Steam with this entropy at 10 kPa is wet, yielding the equation:

$$S_2' = S_2^1 + \kappa_2' (S_2^T - S_2^L)$$

Substituting the values:

6.5597
$$\frac{kJ}{kg \cdot K} = 0.9441 \frac{kJ}{kg \cdot K} + x_2' \left(7.7695 \frac{kJ}{kg \cdot K} - 0.9441 \frac{kJ}{kg \cdot K} \right)$$

 $x_2' = 0.8228$

This is the quality (fraction vapor) of the discharge stream at point 2'. The enthalpy H'_2 is also given by the equation:

$$H_2' = H_2^L + \chi_2' (H_2^p - H_2^L)$$

Thus,

Chemical Engineering Thermodynamics Problems And Solutions

Stanley I. Sandler

Chemical Engineering Thermodynamics Problems And Solutions:

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