You must show all your calculation set-ups to receive full credit. You must report all answers to the correct # of significant figures (sig. figs.).

- 1. Who used a cathode ray tube in his/her discovery of the electron?
 - Robert Millikan
 - Ernest Rutherford

 - (d)J.J. Thomson
 - 2. How many significant figures are in each of the following measurements?
 - a. 0.0017 g 250 les

b. 1390 mL 3 50

c. 4.030 km 4 sea fies

- 3. Which unit represents a microgram?

 - (9) Ha
- The most unreactive group in the Periodic Table is/are the
 - 'ar alkaline-earth metals
 - by metalloids
 - halogens
 - d.) noble gases
- 5. The empirical formula of a compound of carbon and hydrogen is CH₂₅ The molar mass is 28.06 g/mol. What is the molecular formula?
 - (a) C.H.

4-2 (Not

- D. C.H. C. C.H.

- d. CaHee
- 6. Label each of the following elements as metal. nonmetal, or metalloid.
 - Americal
 - b. beryllium _ He lal_
 - c. silicon (Non metra)
 - d. boron __ then melsi
- 7. Write both the name and the symbol of the element in group 13 (3A) and period 3.

Aluminum

Calculate the percentage composition by mass (mass percent) of just the hydrogen in C₂H_a.

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