

3	A little of the sample is treated with dil. HCl acid	Colourless, odourless gas (CO_2) is evolved with effervescence.	Presence of CO_3^{2-} ion
4	A little of the sample is treated with con. H_2SO_4 .	White fuming, pungent smelling gas (HCl) produces dense white fumes when mouth of NH_4OH bottle is shown near the mouth of the test tube. Note: Warm/Heat if necessary	Presence of Cl^- ion
5	A pinch of salt is heated with a few drops of conc. H_2SO_4 for 10-15 seconds.	Reddish brown gas (NO_2) on boiling the mixture with copper turnings and a few drops of conc. H_2SO_4	Presence of NO_3^- ion.

CONFIRMATORY TESTS FOR ANIONS

6	An aqueous solution of the sample is acidified with dil.HCl followed by the addition of BaCl_2 solution.	A white precipitate (BaSO_4) formed, which is insoluble in con.HCl	Presence of SO_4^{2-} ion
7	An aqueous solution of the sample is acidified with dilute acetic acid followed by the addition of lead acetate [$(\text{CH}_3\text{COO})_2\text{Pb}$]	A white precipitate (PbSO_4) formed, which is soluble in Ammonium acetate [$\text{CH}_3\text{COONH}_4$] solution.	SO_4^{2-} ion is confirmed.

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