



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each term with its best definition. Write the letter of the definition that best matches the term.

Column A

1. a. electron transfer
2. a. group compound
3. a. ionic bonding
4. a. ionic bond
5. a. ionic compound
6. a. lattice energy
7. a. lattice
8. a. crystal
9. a. crystal unit
10. a. ionic bonding
11. a. ionic bond
12. a. ionic compound
13. a. lattice energy
14. a. lattice
15. a. crystal
16. a. crystal unit

Column B

1. a bond formed by the transfer of electrons from one atom to another
2. the electrostatic attraction between oppositely charged ions in a crystal
3. the electrostatic attraction between oppositely charged ions in a crystal
4. the electrostatic attraction between the positive and negative ions in a crystal of an ionic compound
5. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
6. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
7. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
8. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
9. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
10. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
11. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
12. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
13. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
14. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
15. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
16. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound

B. Multiple Choice

Choose the letter that best describes each statement.

1. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
2. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
3. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
4. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
5. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
6. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
7. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
8. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
9. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
10. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
11. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
12. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
13. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
14. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
15. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound
16. a. the energy change when one mole of gaseous ions combine to form one mole of solid ionic compound

Chapter 7 Ionic And Metallic Bonding Answer Key

Yan Bai



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