

62 Test - The Chemistry of Life

BIO 1A

- ____ 1. All matter is composed of. a) atoms b) cells c) carbon d) hydrogen
- ____ 2. The periodic table displays _____ arranged by their atomic number.
a) molecules b) protons c) elements d) neutrons
- ____ 3. H₂O is an example of a(n): a) element b) subatomic particle c) atom d) compound
- ____ 4. Which particles are found within the nucleus of an atom?
a) protons and electrons b) protons and neutrons
c) electrons and neutrons d) protons, electrons, and neutrons
- ____ 5. A gecko's feet stick to surfaces due to:
a) cohesion b) adhesion c) hydrogen bonds d) van der Waals forces
- ____ 6. What is responsible for the "stickiness" of water that results in cohesion and adhesion?
a) covalent bonds b) ionic bonds c) hydrogen bonds d) van der Waals forces
- ____ 7. Polar molecules, like water, have what characteristic?
a) partial negative and positive charges
b) multiple hydrogen bonds
c) an uneven number of protons and electrons
d) long chains of carbon
- ____ 8. A water strider can move across the surface of water due to what property of water?
a) adhesion b) cohesion c) polarity d) heat capacity
- ____ 9. Most enzymes found in the body work best at about 40 degrees. This represents the enzyme's:
a) base rate b) substrate c) optimal pH d) optimal temperature
- ____ 10. What is pH?
a) a measurement of how acidic or basic a substance is
b) the optimal temperature enzymes operate at
c) the tendency of water to attract other water molecules
d) the tendency of water to resist changes in state
- ____ 11. Substances that speed up chemical reactions in the body are called:
a) substrates b) enzymes c) carbohydrates d) ions
- ____ 12. What type of bond results from the sharing of electrons?
a) ionic b) carbonated c) covalent d) van der Waals

Chapter Test Chemistry Of Life Answer Key

Albert A Gayle



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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993 **Chemistry and Life** John W. Hill, 1997

Life William K. Purves, 2001 Authoritative thorough and engaging Life The Science of Biology achieves an optimal

balance of scholarship and teachability never losing sight of either the science or the student The first introductory text to present biological concepts through the research that revealed them Life covers the full range of topics with an integrated experimental focus that flows naturally from the narrative This approach helps to bring the drama of classic and cutting edge research to the classroom but always in the context of reinforcing core ideas and the innovative scientific thinking behind them Students will experience biology not just as a litany of facts or a highlight reel of experiments but as a rich coherent discipline

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Structure & Function of the Body Gary A. Thibodeau, Kevin T. Patton, 2000 Structure and Function of the Body 11 The 11

Psychology in Everyday Life (High School) David G. Myers, C. Nathan DeWall, 2017-01-27 The psychology course is often a student favorite because it is so relevant to our everyday lives And like all Myers and DeWall texts *Psychology in Everyday Life* delivers an enjoyable and entertaining read complete with the latest research and an eye catching design Unlike other Myers DeWall books this is a brief offering that makes no assumptions about a student's background or experiences This book is written with the diversity of your students in mind **Study Guide for Life on Earth** Carl S. Pike, 1973 **Inquiry Into**

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ENC Focus, 2000 *The Human Body in Health & Disease* Gary A. Thibodeau, Kevin T. Patton, 2002 *The Human Body in Health Disease* 3rd edition not only presents a body systems approach with a strong emphasis on vocabulary and basic anatomy and physiology concepts but it also presents the basic mechanisms of disease and pathologic conditions associated with each body system This comprehensive text is dominated by two unifying themes the complementarity of structure and function and homeostasis The integrating principle of homeostasis is used to show how normal structure and function is achieved and maintained Failures of homeostasis are shown as basic mechanisms of disease As a result anatomy physiology and pathology emerge as living and dynamic topics of personal interest and relevance to the student at this level Content covers topics in a very friendly writing style that has won the loyalty of many users and has made *Health Disease* one of the top-selling books in this market

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