

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ lattice energy _____	a. ionic bond
_____ bond energy _____	b. covalent bond
_____ enthalpy of vaporization _____	c. metallic bond

2. Use the electronegativity values shown in **Figure 20**, on page 161 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ ionic _____	a. H—F	_____ nonpolar covalent _____	d. H—H
_____ ionic _____	b. Na—Cl	_____ polar covalent _____	e. H—C
_____ polar covalent _____	c. H—O	_____ polar covalent _____	f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

A hydrogen bond is a dipole-dipole attraction between a partially positive hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F. Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction. Hydrogen bonds are generally intermolecular, while ionic and covalent bonds occur between ions or atoms respectively.

4. H₂S and H₂O have similar structures and their central atoms belong to the same group. Yet H₂S is a gas at room temperature and H₂O is a liquid. Use bonding principles to explain why this is.

Oxygen has higher electronegativity than sulfur, which creates a highly polar bond. Increased polarity in H₂O bonds means a stronger intermolecular attraction, making water a liquid at room temperature. Hydrogen bonding exists between water molecules, but not between hydrogen sulfide molecules.

Chapter 8 Covalent Bonds Answers

Julie R. Frentrup



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Answer Book for Chemistry, Sixth Edition Charles E. Mortimer, 1986 **Chemistry** John W. Moore, Conrad L. Stanitski, Peter C. Jurs, 2002 CHEMISTRY THE MOLECULAR SCIENCE is intended to help students develop a broad overview of chemistry and chemical reactions an understanding of the most important concepts and models that chemists and those in chemistry related fields use an appreciation of the many ways chemistry impacts our daily lives the ability to apply the facts concepts and models of chemistry appropriately to new situations in chemistry other sciences and engineering and to other disciplines **A Level Chemistry MCQ (Multiple Choice Questions)** Arshad Iqbal, 2019-06-18 The A Level Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF A Level Chemistry MCQ PDF Download Quiz Questions Chapter 1 28 Practice Tests with Answer Key IGCSE GCE Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs A Level Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests A Level Chemistry MCQ PDF book helps to practice test questions from exam prep notes The A Level Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs A Level Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements tests for college and university revision guide A Level Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book IGCSE GCE Chemistry MCQs Chapter 1 28 PDF includes high school question papers to review practice tests for exams A Level Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam A Level Chemistry Mock Tests Chapter 1 28 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Alcohols and Esters MCQ Chapter 2 Atomic Structure and Theory MCQ Chapter 3 Benzene Chemical Compound MCQ Chapter 4 Carbonyl Compounds MCQ Chapter 5 Carboxylic Acids and Acyl Compounds MCQ Chapter 6 Chemical Bonding MCQ Chapter 7 Chemistry of Life MCQ Chapter 8 Electrode Potential MCQ Chapter 9 Electrons in Atoms MCQ Chapter 10 Enthalpy Change MCQ Chapter 11 Equilibrium MCQ Chapter 12 Group IV MCQ Chapter 13 Groups II and VII MCQ Chapter 14 Halogenoalkanes MCQ Chapter 15 Hydrocarbons MCQ Chapter 16 Introduction to Organic Chemistry MCQ Chapter 17 Ionic Equilibria MCQ Chapter 18 Lattice Energy MCQ Chapter 19 Moles and Equations MCQ Chapter 20 Nitrogen and Sulfur MCQ Chapter 21 Organic and Nitrogen

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Table of Contents Chapter 8 Covalent Bonds Answers

1. Understanding the eBook Chapter 8 Covalent Bonds Answers
 - The Rise of Digital Reading Chapter 8 Covalent Bonds Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 8 Covalent Bonds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 8 Covalent Bonds Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 8 Covalent Bonds Answers
 - Personalized Recommendations
 - Chapter 8 Covalent Bonds Answers User Reviews and Ratings
 - Chapter 8 Covalent Bonds Answers and Bestseller Lists

5. Accessing Chapter 8 Covalent Bonds Answers Free and Paid eBooks
 - Chapter 8 Covalent Bonds Answers Public Domain eBooks
 - Chapter 8 Covalent Bonds Answers eBook Subscription Services
 - Chapter 8 Covalent Bonds Answers Budget-Friendly Options
6. Navigating Chapter 8 Covalent Bonds Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 8 Covalent Bonds Answers Compatibility with Devices
 - Chapter 8 Covalent Bonds Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 8 Covalent Bonds Answers
 - Highlighting and Note-Taking Chapter 8 Covalent Bonds Answers
 - Interactive Elements Chapter 8 Covalent Bonds Answers
8. Staying Engaged with Chapter 8 Covalent Bonds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 8 Covalent Bonds Answers
9. Balancing eBooks and Physical Books Chapter 8 Covalent Bonds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 8 Covalent Bonds Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 8 Covalent Bonds Answers
 - Setting Reading Goals Chapter 8 Covalent Bonds Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 8 Covalent Bonds Answers
 - Fact-Checking eBook Content of Chapter 8 Covalent Bonds Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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