9

CHEMICAL NAMES AND FORMULAS

Chapter Test A

A. Matching

Match each description in Column B to the correct term in Column A. Write the letter of the correct description in the blank.

	Column A		Column B
	. cation	а.	any atom or group of atoms with a negative charge
2	. anion	b.	When two elements combine to form more than one compound, the different masses of one element that combine with the same mass of the other element are in the ratio of small whole numbers.
3	Law of definite proportions	e.	tightly bound group of atoms that behaves as a unit and carries a charge
	. acid	d.	a compound that produces hydroxide ions when dissolved in water
5	. base	e.	a compound composed of two different elements
6	law of multiple proportions	r.	any atom or group of atoms with a positive charge
7	. polyatomic ion	g.	In all samples of the same chemical compound, the elements are always combined in the same proportion by mass.
8	L monatomic ion	h.	a compound that produces hydrogen ions when dissolved in water
9	. binary compound	36.	consists of a single atom with a positive or negative charge

B. Multiple Choice

Choose the best answer and write its letter on the line.

- ______10. The correct name for the N³⁻ ion is the: a. nitrate ion.
 - b. nitric ion.

- c. nitride ion.
- d. nitrite ion.

Chapter 9 Chemical Names And Formulas Test B Answers

James E. Brady, John R. Holum

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