

Part I: Multiple Choice: Write the letter corresponding to the best choice in the space provided.

___C___ 1. In balancing a chemical equation, which of the following are you allowed to do?

- a. change subscripts
- b. change superscripts
- c. add coefficients
- d. add new substances

___C___ 2. What number should be written in front of Na to balance the equation:
$$\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$$

- a. 1
- b. 3
- c. 2
- d. 4

___D___ 3. What is the standard form for a single-replacement reaction?

- a. $\text{AX} + \text{BY} \rightarrow \text{AY} + \text{BX}$
- b. $\text{A} + \text{B} \rightarrow \text{AB}$
- c. $\text{AB} \rightarrow \text{A} + \text{B}$
- d. $\text{A} + \text{BX} \rightarrow \text{AX} + \text{B}$

___D___ 4. How many atoms of oxygen are represented in $2 \text{Ca}(\text{NO}_3)_2$?

- a. 4
- b. 10
- c. 6
- d. 12

Chemistry Chapter 12 Test Answers

Michael Smith



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