

Q 11: Define thermoplastics and thermosetting polymers with two examples of each.

Solution:

Thermosetting polymers are called polymers which are cross-linked or strongly branched polymers which are hardened during the moulding process. Through heating, they cannot be softened yet again.
Eg: urea-formaldehyde resins, bakelite.

Thermoplastic polymers are called polymers which are linear slightly branched long-chain polymers which can be softened and hardened repeatedly when heated.

Eg: polystyrene, polythene.

Q 12: Write the monomers used for getting the following polymers.

(i) Polyvinyl chloride

(ii) Teflon

(iii) Bakelite

Solution:

(1) Vinyl chloride ($CH_2 = CHCl$)

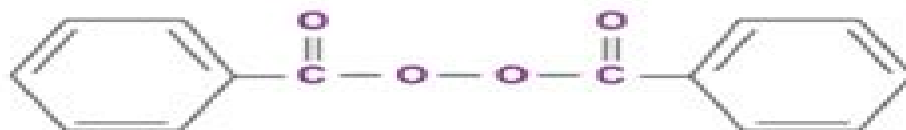
(2) Tetrafluoroethylene ($CF_2 = CF_2$)

(3) Phenol (C_6H_5OH) and Formaldehyde (HCHO)

Q 13: Write the name and structure of one of the common initiators used in free radical addition polymerisation.

Solution:

Benzoyl peroxide is one of the common initiator's name and structure used in free radical polymerization.



Q 14: How does the presence of double bonds in rubber molecules influence their structure and reactivity?

Solution:

Natural rubber is a linear cis-polyisoprene consisting of double bonds between the two units of isoprene C_2 and C_3 .

Chemistry Chapter 15 Solutions Notes

Mark Hyman



Chemistry Chapter 15 Solutions Notes:

Lecture Notes On Solution Chemistry Viktor Gutmann, 1995-08-31 This book emphasises those features in solution chemistry which are difficult to measure but essential for the understanding of both the qualitative and the quantitative aspects Attention is paid to the mutual influences between solute and solvent even at extremely small concentrations of the former The described extension of the molecular concept leads to a broad view not by a change in paradigm but by finding the rules for the organizations both at the molecular and the supermolecular level of liquid and solid solutions **NEET UG Chemistry Paper Study Notes | Chapter Wise Note Book For NEET Aspirants | Complete Preparation Guide with Self Assessment Exercise** EduGorilla Prep Experts, 2022-09-15 Best Selling Book in English Edition for NEET UG Chemistry Paper Exam with objective type questions as per the latest syllabus Increase your chances of selection by 16X NEET UG Chemistry Paper Study Notes Kit comes with well structured Content Chapter wise Practice Tests for your self evaluation Clear exam with good grades using thoroughly Researched Content by experts **Course Notes on the Interpretation of Infrared and Raman Spectra** Dana W. Mayo, Foil A. Miller, Robert W. Hannah, 2004-05-24 Feste fl ssige oder Dampfphase reiner Stoff oder L sung Die IR Spektroskopie ist mittlerweile auf Proben aller Art anwendbar und die Probenmenge darf im Pikogrammbereich liegen Wie man insbesondere IR und Raman Spektren gro er Molek le auswertet und interpretiert zeigt dieses in seiner Art einmalige Werk das als Arbeitsanleitung und Nachschlagewerk gleicherma en geeignet ist An vielen Beispielen kann der Leser sich in der Interpretation von Spektren ben Im Anhang findet sich eine ausf hrliche Bibliographie ansprechend geordnet nach 14 Spezialgebieten **CUET PG Nanoscience Integrative Biosciences Notes** Mocktime Publication, CUET PG Nanoscience Integrative Biosciences Notes CUET PG Books PDFs Chapters Topics Study Material Guide Notes CUET PG CUET PG Previous Papers Question Papers Practice Sets Question Bank CUET PG CUET PG Syllabus Exam Pattern Preparation How to Prepare Revision CUET PG **A Textbook of Physical Chemistry** Arthur Adamson, 2012-12-02 A Textbook of Physical Chemistry Second Edition serves as an introductory text to physical chemistry Topics covered range from wave mechanics and chemical bonding to molecular spectroscopy and photochemistry ideal and nonideal gases the three laws of thermodynamics thermochemistry and solutions of nonelectrolytes The kinetics of gas phase reactions colloids and macromolecules and nuclear chemistry and radiochemistry are also discussed This edition is comprised of 22 chapters the first of which introduces the reader to the behavior of ideal and nonideal gases with particular emphasis on the van der Waals equation The discussion then turns to the kinetic molecular theory of gases and the application of the Boltzmann principle to the treatment of molar polarization dipole and magnetic moments the phenomenology of light absorption and classical and statistical thermodynamics The chapters that follow focus on the traditional sequence of chemical and phase equilibria electrochemistry and chemical kinetics in gas phase and solution phase This book also considers wave mechanics and its applications molecular spectroscopy and photochemistry and the excited

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Cradle For Life, The: The Origin, Evolution And Future Of The Environment Frank D Stacey, Jane Helen

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