

11

CHEMICAL REACTIONS

Chapter Test B

A. Matching

Match each term in Column B with the correct description in Column A. Write the letter of the correct term on the line.

Column A

- _____ 1. a reaction in which an element or compound reacts with oxygen, often producing energy in the form of heat or light
- _____ 2. a reaction in which two or more substances react to form a single substance
- _____ 3. an equation that indicates only those particles that actually take part in the reaction
- _____ 4. a substance that speeds up a reaction without being used up
- _____ 5. ions that are not directly involved in a reaction
- _____ 6. a reaction in which a single compound is broken down into two or more products
- _____ 7. an equation in which each side has the same number of atoms of each element
- _____ 8. a reaction in which atoms of an element replace the atoms of a second element in a compound
- _____ 9. a list of metals in order of decreasing reactivity
- _____ 10. a reaction that involves an exchange of positive ions between two compounds

Column B

- a. decomposition reaction
- b. activity series
- c. spectator ions
- d. balanced equation
- e. double-replacement reaction
- f. catalyst
- g. combustion reaction
- h. combination reaction
- i. net ionic equation
- j. single-replacement reaction

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. In the chemical equation $2\text{H}_2\text{O}_2(\text{aq}) \xrightarrow{\text{MnO}_2} 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$, the MnO_2 is a:
- a. reactant.
b. product.
c. spectator ion.
d. catalyst.
- _____ 12. When the equation $\text{Mg}(\text{s}) + \text{HCl}(\text{aq}) \rightarrow \text{MgCl}_2(\text{aq}) + \text{H}_2(\text{g})$ is balanced, what is the coefficient for HCl ?
- a. 1
b. 2
c. 3
d. 4

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Yijin Wang



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Table of Contents Chemistry Chapter 11 Chemical Reactions Test Answers

1. Understanding the eBook Chemistry Chapter 11 Chemical Reactions Test Answers
 - The Rise of Digital Reading Chemistry Chapter 11 Chemical Reactions Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 11 Chemical Reactions Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter 11 Chemical Reactions Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 11 Chemical Reactions Test Answers
 - Personalized Recommendations
 - Chemistry Chapter 11 Chemical Reactions Test Answers User Reviews and Ratings
 - Chemistry Chapter 11 Chemical Reactions Test Answers and Bestseller Lists
5. Accessing Chemistry Chapter 11 Chemical Reactions Test Answers Free and Paid eBooks
 - Chemistry Chapter 11 Chemical Reactions Test Answers Public Domain eBooks
 - Chemistry Chapter 11 Chemical Reactions Test Answers eBook Subscription Services
 - Chemistry Chapter 11 Chemical Reactions Test Answers Budget-Friendly Options

6. Navigating Chemistry Chapter 11 Chemical Reactions Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Chapter 11 Chemical Reactions Test Answers Compatibility with Devices
 - Chemistry Chapter 11 Chemical Reactions Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Chapter 11 Chemical Reactions Test Answers
 - Highlighting and Note-Taking Chemistry Chapter 11 Chemical Reactions Test Answers
 - Interactive Elements Chemistry Chapter 11 Chemical Reactions Test Answers
8. Staying Engaged with Chemistry Chapter 11 Chemical Reactions Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 11 Chemical Reactions Test Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 11 Chemical Reactions Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 11 Chemical Reactions Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 11 Chemical Reactions Test Answers
 - Setting Reading Goals Chemistry Chapter 11 Chemical Reactions Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Chapter 11 Chemical Reactions Test Answers
 - Fact-Checking eBook Content of Chemistry Chapter 11 Chemical Reactions Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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