

14

THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the word.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed mass of gas at constant pressure
2. The volume of a fixed mass of gas at constant pressure when the absolute temperature is 273°K
3. The ratio of the mass of a gas to its volume, independent of the state of the substance
4. Temperature at which the volume of a gas becomes zero
5. A measure of how much the volume of a gas decreases when cooled
6. The a parameter of gas pressure, temperature, and volume of the gas mathematically expressed
7. An ideal gas that obeys the gas laws perfectly
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of the force through a unit surface of a container of gas
10. A unit of volume, 22.4 liters

Column II

- a. standard pressure
- b. ideal gas constant R
- c. density
- d. extrapolation
- e. Boyle's law
- f. Boyle's equation
- g. Charles' law of expansion
- h. diffusion
- i. compressibility
- j. Graham's law of diffusion

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Diffusion of a mixture of gases is dependent on the mass of the gases, which is why, in perfume, the lighter scents spread more rapidly than the heavier ones.
a. the concentration of the gases
b. the volume of the gases
c. the pressure of the gases
d. the density of the gases

Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers

LM Reder



Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers:

Chemistry and Our Universe , In the first of two lectures on the properties of gases review the basic equations that describe their behavior Learn the history of Boyle s law Gay Lussac s law Charles s law and Avogadro s law Then use these four expressions to derive the celebrated ideal gas law

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Table of Contents Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers

1. Understanding the eBook Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - The Rise of Digital Reading Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Personalized Recommendations
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers User Reviews and Ratings
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers and Bestseller Lists
5. Accessing Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers Free and Paid eBooks
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers Public Domain eBooks
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers eBook Subscription Services
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers Compatibility with Devices
 - Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Highlighting and Note-Taking Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Interactive Elements Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
8. Staying Engaged with Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Setting Reading Goals Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Fact-Checking eBook Content of Chemistry Chapter 14 The Behavior Of Gases Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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