

MACROMOLECULES - CARBOHYDRATES

Text, Section 1.2 – The Chemicals of Life, pages 24-34

1. Define the following terms:

a) Macromolecule

A very large molecule, such as protein, commonly created by the combination of smaller subunits (monomers).

b) Monomer

The subunits that make up the macromolecules

c) Polymer

A substance that has a molecular structure consisting mostly or entirely of a large number of similar units (monomers) bonded together; sort of a synonym for macromolecule

2. What are the four major groups of macromolecules that are important to biology?

Carbohydrates, Lipids, Proteins, Nucleic Acids

3. Define carbohydrate in chemical terms.

a) The general formula for a carbohydrate is $C_xH_{2y}O_y$

b) If a sugar has 11 oxygen atoms, how many hydrogen atoms does it contain? 22

c) Based on their molecular formula, which of the following are NOT carbohydrates?

i. $C_3H_5O_3$

ii. $C_{10}H_{19}O_9$

iii. $C_{15}H_{32}O_{15}$

iv. $C_4H_6O_2$

v. $C_{15}H_{32}O_2$

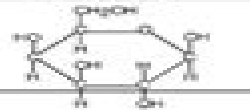
vi. $C_6H_{12}O_6$

THERE ARE THREE TYPES OF CARBOHYDRATES

A. Monosaccharides are the **monomers** of carbohydrates. They are the simple, sweet-tasting sugars.

4. Mono means: **one** The most important simple sugar for humans is GLUCOSE. Glucose is represented by the chemical formula $C_6H_{12}O_6$

Important simple sugars:

Name & Chemical Formula	Description/Use	Diagram	How are they similar	How are they different?
Glucose	Primary energy-storage unit used by cells		All have the same chemical formula	Different arrangement of atoms (i.e they are isomers)
Fructose	Sugar found in fruit			
Galactose	Sugar found in milk (it's one of the monomer units of lactose)			

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