Chemistry of Life notes

Organic Molecules: molecules that contain both Carbon and Hydrogen (C and H); can contain
other elements too; - IF THEY CONTAIN ONLY C AND H then they can ALSO be called
hydrocarbons.

What is so cool about carbon?

Carbon has _four valence electrons which means it can form 4 bonds AND can bond with itself.

- This means it can form long chains, rings, branched chains and many other molecules

Inorganic molecules: molecules that do NOT contain both C and H







WATER: most essential inorganic molecule

- ♦ Body is 65- 75% water on average
- More water in fat or muscle?? MUSCLE Why? Fat is hydrophobic
- More water in Males or females?? Males Why? Males have less fat than females

Functions of water:

- Solvent: dissolves lots of stuff (salts especially)
- Medium: where the chemical reactions happen
- Moistens Surfaces: lungs, mouth, nasal passages
- Temperature Regulation: an example is sweating (keeps internal temperature from getting too high)
- Cushion Brain in skull;
- Transportation: moves nutrients and other materials through the body (blood is mostly water!)
- · Lubrication: in your Joints; prevent bones from scraping
- Hydrolysis: type of chemical reaction that uses water to break apart large molecules

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Katherine Blood Hoffman

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Lectures on the Chemistry of Life Thomas Turner, Frederick Crace Calvert, 1847 The Chemistry of Life's Origins I Mayo Greenberg, C X Mendoza-Gomez, Valerio Pirronello, 1993-10-31 What is Life? Addy Pross, 2012-09-27 Seventy years ago Erwin Schr dinger posed a profound guestion What is life and how did it emerge from non life This problem has puzzled biologists and physical scientists ever since Living things are hugely complex and have unique properties such as self maintenance and apparently purposeful behaviour which we do not see in inert matter So how does chemistry give rise to biology What could have led the first replicating molecules up such a path Now developments in the emerging field of systems chemistry are unlocking the problem Addy Pross shows how the different kind of stability that operates among replicating molecules results in a tendency for chemical systems to become more complex and acquire the properties of life Strikingly he demonstrates that Darwinian evolution is the biological expression of a deeper well defined chemical concept the whole story from replicating molecules to complex life is one continuous process governed by an underlying physical principle The gulf between biology and the physical sciences is finally becoming bridged This new edition includes an Epilogue describing developments in the concepts of fundamental forms of stability discussed in the book and their profound implications Oxford Landmark Science books are must read classics of modern science writing which have crystallized big The Chemistry of Life ideas and shaped the way we think **Chemistry and Life** Meeting on Chemistry and Life,2008 The Tao of Chemistry and Life Eugene H. Cordes, 2009-06-04 Chemistry underlies life This book Robert Thornton.2001 establishes the relationship between the focal point of chemistry the molecule and the key characteristics of living organisms The key is the interactions between small molecules and macromolecules leading to metabolic control memory and learning the senses and drug action The Chemistry of Life Joseph Needham, 1970 Chemistry of Life Peter P. Berlow, 1982

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