

• **States of Matter**

Diffusion

• **Experimental Techniques**

Apparatus

Separation techniques: Filtration, Simple distillation & Fractional distillation, chromatography, centrifugation

Determination of melting point and boiling point

Methods of collecting gases

Drying agents

Reading values from burette, stop watch, measuring cylinder, thermometer, gas syringe etc.

Hazard Symbols

• **Electricity and Chemistry**

Electrolysis of molten electrolytes and aqueous electrolytes

Electroplating

Simple cells

Energy produced when a fuel burns. (Text Book Page 181)

Verify the given change is exothermic or endothermic.

• **Chemical Reactions      (Text Book Page 185-204)**

Describe the effect of concentration, particle size, catalysts (including enzymes) and temperature on the speeds of reactions

Describe a practical method for investigating the speed of a reaction involving gas evolution

Devise a suitable method for investigating the effect of a given variable on the speed of a reaction

Interpret data obtained from experiments concerned with speed of reaction

Describe and explain the effects of temperature and concentration in terms of collisions between reacting particles

To verify the given substance is a catalyst

Experimental set up for the preparation of gases: hydrogen, oxygen, sulphur(VI) oxide, nitrogen dioxide, chlorine

# Chemistry Paper 2 Igcse Grade 9 Exams

**Mary Falkner**



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book also includes many black and white photographs lyrics random thoughts and artwork the book was

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gave each student 20 marshmallows and a handful of toothpicks next i had them make a base by sticking 4 toothpicks into 4 marshmallows and making a square then they could build their sculpture up from there

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