

Stoichiometry Calculation Practice Worksheet

1. Calculate the number of moles of NaOH that are needed to react with 500.0 g of H_2SO_4 according to the following equation:

**ANS: 10.19 mol**

2. Calculate the mass of NH_3 that can be produced from the reaction of 125 g of NCl_3 according to the following equation:

**ANS: 17.7 g**

3. Identify the limiting reactant and determine the mass of CO_2 that can be produced from the reaction of 25.0 g of C_3H_8 with 75.0 g of O_2 according to the following equation:

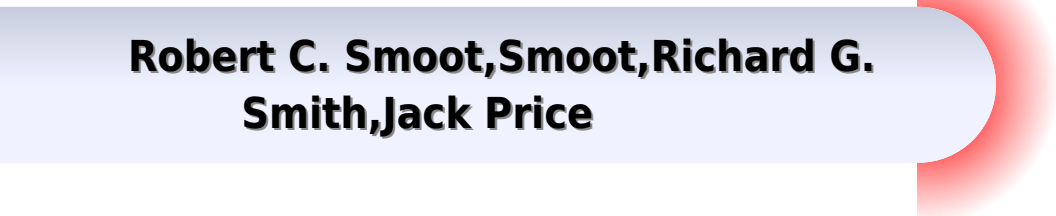
**ANS: 61.9 g**

4. How many grams of SO_2 are produced when 152 g of CS_2 react with 48.0 g of O_2 according to the following equation:

**ANS: 64.1 g**

Chemistry Unit 5 Stoichiometry Practice Problems I

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Smith, Jack Price**



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