

TEST 1

COMMUNICATION MANAGEMENT N4

QUESTION 1: BASIC COMMUNICATION PRINCIPLES

- 1.1 Choose a term from COLUMN B that matches a description in COLUMN A. Write only the letter (A–J) next to the question number (1.1.1–1.1.10) in the ANSWER BOOK.

COLUMN A		COLUMN B	
1.1.1	Hesitation markers like um and ah used during communication	A	mass communication
1.1.2	Body movements like gestures, facial expressions and body posture	B	filler sounds
1.1.3	Distance of over three metres maintained between the sender and audience	C	noise
1.1.4	Umbrella term for communication barriers	D	kinesics
1.1.5	Minister of Health announces the Covid regulations on SABC 2	E	public zone

(5× 1) (5)

- 1.2 Listening is a skill one can learn and perfect to improve communication.

Distinguish between *attentive listening* and *critical listening*.

(2+2)

(4)

- 1.3 What are the elements of the communication process, list and explain them

(6)

Management Communication N4 Question Paper Aehret

Marlene J. Van Schalkwyk, M. J. Viljoen

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Communication and Management Communication N4 SB A Thorne, 2019 *N4 Communication & Management*
Communication Frieda Wade, 2018 Communication for Management and Secretarial Students Marlene J. Van
Schalkwyk, M. J. Viljoen, 1996

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $HCl(aq) + NaOH(aq) \rightarrow NaCl(aq) + H_2O(l)$
one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $H_2C_2O_4 \cdot 2H_2O$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ $4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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