

PVA polymer slime

Class practical

A solution of **polyvinyl alcohol (PVA)** can be made into a **slime** by adding **borax** solution, which creates crosslinks between **polymer** chains. In this activity, some interesting properties of the slime are investigated. Students are guaranteed to enjoy the activities involved.

Lesson organisation

This experiment is easy to set up – providing the chemicals are available, and should take no more than about 30 minutes. It can be done by students in groups of two or three.

Apparatus	Chemicals
Goggles	Each working group requires:
Each working group requires:	PVA, polyvinyl alcohol, 4% (or 8%) aqueous solution, 40 cm ³
Beaker (100 cm ³)	Borax, hydrated sodium tetraborate, 4% (or 8%) aqueous solution, 10 cm ³
Measuring cylinder (50 cm ³)	Food colour or fluorescein (optional)
Disposable plastic cup	Hydrochloric acid, about 0.5 M, 20 cm ³ (optional)
Metal spatula	Sodium hydroxide, about 0.5 M (CORROSIVE), 20 cm ³ (optional)
Petri dish (or watch glass)	Refer to Health & Safety and Technical notes section below for additional information.
Water-based felt-tipped pen	
Spirit-based felt-tipped pen	
Disposable plastic gloves	

Health & Safety and Technical notes

[Read our standard health & safety guidance](#)

Wear eye protection, and protective gloves if handling the slime.

Polyvinyl alcohol, $(-[\text{CH}_2\text{CH}(\text{OH})]_n-)$ - PVA can be high MW (about 120 000) or low MW (about 15 000). If high MW PVA is used, prepare a 4% solution by placing 960 cm³ of water into a tall 1 dm³ beaker. Measure out 40 g of high MW PVA and add this slowly to the beaker of water, with stirring. If low MW PVA is used, prepare an 8% solution by placing 920 cm³ of water into a tall 1 dm³ beaker. Measure out 80 g of low MW PVA and add this slowly to the beaker of water, with stirring. In each case, heat the mixture gently, stirring occasionally, until the solution clears. Avoid boiling the solution. After cooling, this solution can be poured into suitable smaller containers, which can then be sealed and stored indefinitely.

Borax, hydrated sodium tetraborate, $(\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O})$ - see CLEAPSS Hazcard. If a 4% aqueous solution of PVA is used a 4% aqueous solution of borax will be required. If an 8% aqueous solution of PVA is used an 8% aqueous solution of borax will be required. Borax solid is now classified as TOXIC because it carries the R60/61 warning – it may cause infertility and harm to the unborn child. The solid and solutions in excess of 8.5% should be labelled TOXIC, but other solutions will NOT carry any hazard warning. Borax solutions as prepared following these instructions do NOT require a hazard label. BUT technicians preparing the solutions need to be alerted to the hazard. CLEAPSS advises weighing the solid very carefully, possibly in a fume cupboard which is not switched on. Pregnant women should avoid handling the solid. Students may want to take the slime home - this

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**Tabli Ghosh,Ruchir Priyadarshi,Swarup
Roy**



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