

7. Test for Hg_2^{2+} , Co^{2+} , and Al^{3+}

Some 6 M HCl is added to the colorless solution. No precipitate forms. Next, some 3% H_2O_2 and 6 M NaOH are added until the solution is basic, and it is then heated. No precipitate forms. The solution is acidified with 6 M HNO_3 and then made basic with 6 M NH_3 . No precipitate forms.

Ions present: _____ Ions absent: _____ Ions undetermined: _____

8. Test for Ag^+ , Fe^{3+} , Bi^{3+} , and Zn^{2+}

Some 6 M HCl is added to the colorless solution. A white precipitate forms and is removed. The solution is then made basic with 6 M NaOH. No precipitate forms.

Ions present: _____ Ions absent: _____ Ions undetermined: _____

Only one of each of the following pairs of reactants undergoes a reaction. Complete and balance the equation for the reaction which occurs.



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