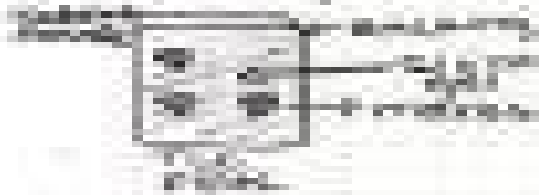


1. Prepare the standards by adding the solvent mixture (methylacetone: CH₂Cl₂) and adding the stock solution.
2. Measure the IR values of the individual components.
3. Add the IR values to the file of the measurement.
4. Identify the components of the mixture.

Example:



IR Spectrum



IR Spectrum

1. IR value
2. IR value
3. IR value
4. IR value
5. IR value
6. IR value
7. IR value
8. IR value
9. IR value
10. IR value

IR Spectrum

$$IR = 1715 \text{ cm}^{-1}$$

$$IR = 1715 \text{ cm}^{-1}$$

$$IR = 1715 \text{ cm}^{-1}$$

Hayden Mcneil Chemistry Lab Answers

Sara Jane Dallas



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Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-02-23 Winner of the CHOICE Outstanding Academic Title 2017 Award. This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education. Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future. Adopting a practice oriented approach, the current challenges and opportunities posed by chemistry education are critically discussed, highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them. The main topics discussed include best practices, project based education, blended learning, and the role of technology, including e learning and science visualization. Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively, from experienced chemistry professors to secondary school teachers, from educators with no formal training in didactics to frustrated chemistry students.

Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging. Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout. Each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives. Essentials of General Organic and Biochemistry captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course. Students value their experience with chemistry, getting a true sense of just how relevant it is to their chosen profession. To browse a sample chapter, view sample ChemCasts, and more, visit www.whfreeman.com/gob.

Exploring Chemistry (Loose-Leaf) Matthew Johll, 2012-05-04 Matthew Johll's Exploring Chemistry covers the standard topics for the nonmajors course in the typical order, but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations. For example, students work through the often difficult topics of

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