

concerned in agriculture. Some of the elements are termed *organic*, because they were supposed, formerly, always to be found in organised beings—that is, in plants and animals; and the rest are called *inorganic*, as they are not necessarily present in living things. But organic elements occur in enormous quantities in inorganic substances, and inorganic elements are found in organised bodies, so that the terms organic and inorganic elements are now used simply as convenient expressions.

Organic elements.

Inorganic elements.

The organic elements are *oxygen*, *hydrogen*, *nitrogen* and *carbon*, which disappear as vapour or smoke when an organised body is burnt in the air. The first three are gases, and the last is a substance, one form of which, charcoal, is familiar to every one. Besides these, there are two other substances, namely, *sulphur* and *phosphorus*, which are sometimes called secondary organic elements, because they are frequently found in organised bodies. In all, then, there are six non-metals, as they are termed, ever present in fertile soils, and, in addition to them, there are five metals which are always to be found in combination with other elements, in lands suitable for cultivation. These are *potassium*, *sodium*, *magnesium*, *calcium* and *iron*.

Elements in the soil.

Eleven secondary elements.

Oxygen is an invisible gas which enters largely into the composition of the atmosphere; and in combination with another gas, hydrogen, it forms water. It combines readily with other elements to form different compounds called *oxides*. If iron be exposed to air and moisture its surface rusts, and this rust is simply a combination of iron and the gas oxygen; it is in fact an *oxide of iron*. Oxygen also combines with carbon to form *carbon dioxide*, usually called *carbonic acid*, a compound gas that plays a very important part in plant life. In combination with the element silicon, oxygen forms the oxide *silica*, which is the basis of all sands; with potassium it forms *potash*; with sodium it forms *soda*; with magnesium it forms *magnesia*; with calcium it forms *lime*, and with aluminium it forms *alumina*, which is the basis of all clays.

Compounds of oxygen.

Oxides.



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