#### **SAMPLE PAPER (2023 -24)**

#### CHEMISTRY THEORY (043)

Max. Marks:70

Time: 3 hours

General Instructions:

Read the following instructions carefully.

- (a) There are 33 questions in this question paper with internal choice.
- (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
- (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
- (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
- (e) SECTION D consists of 2 case based questions carrying 4 marks each.
- (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
- (q) All questions are compulsory.
- (h) Use of log tables and calculators is not allowed.

#### SECTION A

The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.

- 1. Which of the following solutions will have the highest conductivity at 298 K?
  - (a) 0.01 M HCl solution

(b) 0.1 M HCl solution

(c) 0.01 M CH 3COOH solution

(d) 0.1 M CH 3COOH solution

Identify A and B:

- (a) A = 1-phenylethanal, B = acetophenone
- (b) A = Benzophenone B = formaldehyde
- (c) A= Benzaldehyde, B = Acetophenone
- (d) A = Benzophenone, B = Acetophenone
- The vitamins which can be stored in our body are:
  - (a) Vitamin A, B, D and E

(d) Vitamin A, C, D and K

(c) Vitamin A, B, C and D

- (d) Vitamin A, D, E and K
- 4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give

CH3 CH= C(CH3)COONa and yellow precipitate of CHI3?

(a) 3-Methylpent-3-en-2one

(b) 3-Methylbut-2-en- one

(c) 2, 3-Dimethylethanone

(d) 3-Methylpent-4-one

- 5. Which of the following is not correct?
  - (a) In haloarenes, the electron pairs on halogen atom are in conjugation with  $\pi$ -electrons of the ring.
  - (b) The carbon-magnesium bond is covalent and non-polar in nature .
  - (c) During S<sub>N</sub>1 reaction, the carbocation formed in the slow step being sp2 hybridised is planar.
  - (d) Out of CH2= CH-Cl and C6H3CH2Cl, C6H3CH2Cl is more reactive towards SN1 reaction

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## Table of Contents Design Of The Question Paper Chemistry Class Xii Cbse

- 1. Understanding the eBook Design Of The Question Paper Chemistry Class Xii Cbse
  - o The Rise of Digital Reading Design Of The Question Paper Chemistry Class Xii Cbse
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Design Of The Question Paper Chemistry Class Xii Cbse
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Design Of The Question Paper Chemistry Class Xii Cbse
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Design Of The Question Paper Chemistry Class Xii Cbse
  - Personalized Recommendations
  - o Design Of The Question Paper Chemistry Class Xii Cbse User Reviews and Ratings
  - Design Of The Question Paper Chemistry Class Xii Cbse and Bestseller Lists
- 5. Accessing Design Of The Question Paper Chemistry Class Xii Cbse Free and Paid eBooks
  - Design Of The Question Paper Chemistry Class Xii Cbse Public Domain eBooks
  - Design Of The Question Paper Chemistry Class Xii Cbse eBook Subscription Services
  - o Design Of The Question Paper Chemistry Class Xii Cbse Budget-Friendly Options

- 6. Navigating Design Of The Question Paper Chemistry Class Xii Cbse eBook Formats
  - o ePub, PDF, MOBI, and More
  - Design Of The Question Paper Chemistry Class Xii Cbse Compatibility with Devices
  - Design Of The Question Paper Chemistry Class Xii Cbse Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Design Of The Question Paper Chemistry Class Xii Cbse
  - Highlighting and Note-Taking Design Of The Question Paper Chemistry Class Xii Cbse
  - o Interactive Elements Design Of The Question Paper Chemistry Class Xii Cbse
- 8. Staying Engaged with Design Of The Question Paper Chemistry Class Xii Cbse
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - $\circ$  Following Authors and Publishers Design Of The Question Paper Chemistry Class Xii Cbse
- 9. Balancing eBooks and Physical Books Design Of The Question Paper Chemistry Class Xii Cbse
  - Benefits of a Digital Library
  - o Creating a Diverse Reading Collection Design Of The Question Paper Chemistry Class Xii Cbse
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Design Of The Question Paper Chemistry Class Xii Cbse
  - Setting Reading Goals Design Of The Question Paper Chemistry Class Xii Cbse
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Design Of The Question Paper Chemistry Class Xii Cbse
  - Fact-Checking eBook Content of Design Of The Question Paper Chemistry Class Xii Cbse
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements

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