

Diffusion and Osmosis Worksheet

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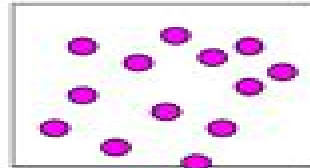
Diffusion is the movement of particles from areas of higher concentration to areas of lower concentration. It is a natural, random process. This means that it does not require extra energy input.

1. These are pictures of molecules frozen at two different times. Draw arrows to show the direction each particle might travel due to diffusion in diagrams A and B.

A.



B.



2. Diagram A shows the same number of particles as in Diagram B except most particles start out on one side of the box. Explain why after a while, Diagram A resembles Diagram B.

3. Give an everyday example of diffusion in air and in water.

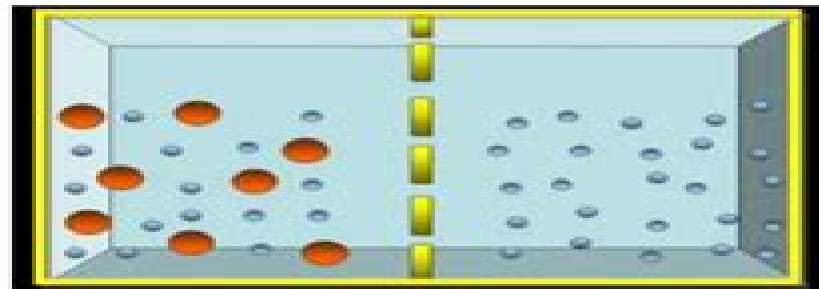
- AIR:
- WATER:

WORD BANK: Passive / Active / Lower / Higher / Unevenly / Evenly

4. Complete the following sentence: Diffusion is the _____ movement of particles from _____ concentration to a _____ concentration until they are _____ spread out.
5. Moths emit chemicals called pheromones to attract a mate.
- a) Which process is responsible for the distribution of these chemicals through the air?
 - b) If a moth detects pheromones, how might it work out which direction they are coming from?
 - c) What would be the advantage of a moth releasing the pheromones on a night when there was little wind?
6. Osmosis is the diffusion of water across a semi permeable membrane. The semi permeable membrane acts like a filter that lets only the water through. Water always goes from the area of higher water concentration to the area of lower water concentration. Confused? Then imagine the particles (or solute) are salt and follow the rule - SALT SUCKS!

On the diagram to the right label the:

- Semi permeable membrane (M)
- Area of higher water concentration (HC)
- Area of lower water concentration (LC)
- Direction of osmotic flow (arrow)



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1. Understanding the eBook Diffusion Osmosis Questions And Answers
 - The Rise of Digital Reading Diffusion Osmosis Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Diffusion Osmosis Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Diffusion Osmosis Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Diffusion Osmosis Questions And Answers
 - Personalized Recommendations
 - Diffusion Osmosis Questions And Answers User Reviews and Ratings
 - Diffusion Osmosis Questions And Answers and Bestseller Lists

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 - Highlighting and Note-Taking Diffusion Osmosis Questions And Answers
 - Interactive Elements Diffusion Osmosis Questions And Answers
8. Staying Engaged with Diffusion Osmosis Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Diffusion Osmosis Questions And Answers
9. Balancing eBooks and Physical Books Diffusion Osmosis Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Diffusion Osmosis Questions And Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Diffusion Osmosis Questions And Answers
 - Setting Reading Goals Diffusion Osmosis Questions And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Diffusion Osmosis Questions And Answers
 - Fact-Checking eBook Content of Diffusion Osmosis Questions And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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