

Chemical Bonding

Section 1: Introduction to Chemical Bonding

In this section, you will explore the basics of chemical bonding and its importance in understanding molecular structure and properties.

1. Which of the following is a covalent bond? (Select all that apply.)

- a. H_2
- b. $NaCl$
- c. CO_2
- d. CH_4

2. In which molecule is the bond length the longest?

- a. H_2
- b. Cl_2
- c. Br_2
- d. I_2

3. Which of the following is a polar covalent bond? (Select all that apply.)

Bond	Electronegativity Difference	Polarity	Dipole Moment
$H-F$	1.9	Yes	Large
$H-Cl$	0.9	Yes	Small

- a. $H-F$
- b. $H-Cl$
- c. $H-Br$
- d. $H-I$

4. Which of the following is a non-polar covalent bond? (Select all that apply.)

Bond	Electronegativity Difference	Polarity	Dipole Moment
$H-H$	0.0	No	None
$Cl-Cl$	0.0	No	None

- a. $H-H$
- b. $H-Cl$
- c. $Cl-Cl$
- d. $H-Br$

Section Quiz Introduction To Chemical Bonding Answers

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Section Quiz Introduction To Chemical Bonding Answers:

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