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Peter,2005-01-01 Peugeot 305 Owners Workshop Manual R. G. O. Hawes,1989 Peugeot Diesel Engine Service and

Repair Manual A. K. Legg,Finn Deacon,1996 PEUGEOT 205 PETROL 1983-1997AP,2019-08-12 Peugeot 205 Owner's

Workshop Manual Andrew Kirk Legg,1995 Peugeot 106 Service and Repair Manual Mark Coombs,Steve Rendle,1999

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Peugeot 404 1960-75 Owners Workshop Manual Veloce Press,2001-11-01 VelocePress in close cooperation with Brooklands Books Ltd has brought this and other repair manuals previously published as part of the Autobooks Owners Workshop Manual Series back into print The series is an invaluable resource for the classic car enthusiast and a must have for owners interested in performing their own maintenance Peugeot 405 (diesel) Service and Repair Manual Steve Rendle,1996

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for the nitration $4\,01\,$ ml of nitric acid $4\,00\,$ ml of sulfuric acid and $4\,56\,$ ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only $4\,56\,$ ml of bromobenzene was used the amount of product predicted to be recovered was $8\,7533\,$ g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer experiment 11 electrophilic aromatic substitution nitration - Jan 12 2022

on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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