

स्मरणीय बिंदु -

- परमाणु विद्युत का उदासीन होता है।
- थॉमसन मॉडल के अनुसार, परमाणु का समस्त द्रव्यमान लगभग 10-10 मीटर बिज्या के ठोस गोले में सीमित रहता है। तथा इसी में इलेक्ट्रॉन समान रूप में वितरित एवं स्थिर रहते हैं। इसे 'प्लम-पुडिंग मॉडल' भी कहा गया।
- रदरफोर्ड के प्रयोग से यह स्पष्ट हुआ की नाभिक में धन आवेश इकट्ठा रहता है, तथा इलेक्ट्रॉन नाभिक के चारों ओर घबकर लगाते हैं। इसे 'वर्णीय मॉडल' या 'नाभिकीय मॉडल' भी कहा जाता है।
- रदरफोर्ड के परमाणु मॉडल के अनुसार,
 - परमाणु उदासीन होता है।
 - परमाणु का समस्त द्रव्यमान और समस्त धनावेश उसके केन्द्र में बहुत ही छोटे भाग में, (नाभिक में) संकेंद्रित रहते हैं।
 - इलेक्ट्रॉन नाभिक के चारों ओर किसी भी कक्षा में घूम सकते हैं।
- रदरफोर्ड के परमाणु मॉडल के दो दोष थे-
 - नाभिक के चारों ओर घूमते हुये त्वरित इलेक्ट्रॉनों को सर्पिलाकार गति करते हुए नाभिक की ओर चले जाना चाहिये। यह पदार्थ के स्थायित्व का खंडन करता है।
 - यह विभिन्न तत्वों के परमाणुओं के अभिलाक्षणिक लाईन स्पेक्ट्रम की व्याख्या नहीं करता।
- नाभिक के पास जहाँ तक ऐल्फा कण पहुंच जाते हैं, उसे 'पहुंच की निकटतम दूरी' कहते हैं।

$$\text{इसका मान } r_0 = \frac{1}{4\pi\epsilon_0} \cdot \frac{4Ze^2}{mv^2} = \frac{1}{4\pi\epsilon_0} \cdot \frac{2E}{K}$$

जहाँ, Ze = नाभिक पर आवेश, m = ऐल्फा कण का द्रव्यमान, v = ऐल्फा कण का प्रारम्भिक वेग तथा $K = \frac{1}{2}mv^2$ ऐल्फा-कण की गतिज ऊर्जा

- किसी इलेक्ट्रॉन की चाल उसके मुख्य क्वांटम नम्बर के व्युत्क्रमानुपाती होती है।
- जब ऐल्फा-कण परमाणु से बहुत अधिक दूरी पर होते हैं, तब नाभिक के केन्द्र से ऐल्फा-कण के वेग सदिश के बीच की दूरी को ऐल्फा-कण का संघट्ट प्राचल कहते हैं। इसे b से व्यक्त करते हैं। $b = \frac{1}{4\pi\epsilon_0} \cdot \frac{2Ze^2 \cdot \cos(\frac{\theta}{2})}{mv^2} = \frac{1}{4\pi\epsilon_0} \cdot \frac{2e^2 \cdot \cos(\frac{\theta}{2})}{K}$
- $K = \frac{1}{2}mv^2$ = ऐल्फा-कण की गतिज ऊर्जा है जब वह परमाणु से बहुत दूर होता है।
- किसी परमाणु के नाभिक से ऐल्फा-कण का संघट्ट प्राचल पर निर्भर करता है।
 - यदि संघट्ट प्राचल b का मान अधिक हो तो ऐल्फा-कण अविचलित हुए चले जाते हैं।
 - यदि संघट्ट प्राचल b का मान कम हो तो ऐल्फा-कण अधिक कोण से प्रकीर्णित हो जाते हैं।
 - यदि संघट्ट प्राचल $b = 0$ हो तो $\cos(\theta/2) = 0$ तब $\theta = 180^\circ$ इस प्रकार, जो ऐल्फा-कण नाभिक के केन्द्र की ओर गति करते हैं उनके लिये संघट्ट प्राचल का मान शून्य होता है ऐसी स्थिति में ऐल्फा-कण प्रतिक्षिप्त (rebound) हो जाते हैं।
- प्रत्येक तत्व के परमाणु स्थायी होते हैं और अभिलाक्षणिक स्पेक्ट्रम उत्पन्न करते हैं। स्पेक्ट्रम अलग-अलग समांतर रेखाओं का समुच्चय होता है, जिसे रेखिल वर्णक्रम कहते हैं। यह परमाणु संरचना के सम्बन्ध में महत्वपूर्ण सूचना देता है।
- परमाण्विक हाइड्रोजन अनेक श्रेणी युक्त रेखिल स्पेक्ट्रम उत्सर्जित करता है। किसी श्रेणी में किसी रेखा की आवृत्ति को दो पदों के अंतर के रूप में व्यक्त किया जा सकता है।

लाइमन (Lyman) श्रेणी: $\nu = Rc \left(\frac{1}{2^2} - \frac{1}{n^2} \right); n = 3, 4, \dots$

बाल्मर (Balmer) श्रेणी: $\nu = Rc \left(\frac{1}{2^2} - \frac{1}{n^2} \right); n = 3, 4, 5, \dots$

पाशन (Paschen) श्रेणी: $\nu = Rc \left(\frac{1}{3^2} - \frac{1}{n^2} \right); n = 4, 5, 6, \dots$

ब्रेकेट (Brackett) श्रेणी: $\nu = Rc \left(\frac{1}{4^2} - \frac{1}{n^2} \right); n = 5, 6, 7, \dots$

फुंड (Pfund) श्रेणी: $\nu = Rc \left(\frac{1}{5^2} - \frac{1}{n^2} \right); n = 6, 7, 8, \dots$

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