

- (c) 350 cm, convergent
(d) 150 cm, divergent
16. Three resistances A, B and C each of 5Ω are connected to a battery of 10 V as shown in the figure. Calculate the current through C:
- 
- (a) 2 amp (b) 1 amp
(c) 3 amp (d) 0.6 amp
17. The device used to generate electrical energy is:
- (a) Electric motor (b) Generator
(c) Galvanometer (d) Voltmeter
18. The number of molecules in 11 g of CO_2 are:
- (a) 0.25×10^{23} (b) 0.50×10^{23}
(c) 1.00×10^{23} (d) 1.51×10^{23}
19. The value of charge / mass ratio of electron was determined by:
- (a) W.K. Roentgen (b) J.J. Thomson
(c) Marie Curie (d) Niels Bohr
20. Which of the following reactions is an example of combination reaction?
- (a) $\text{Zn(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Cu(s)}$
(b) $\text{BaCl}_2(\text{aq}) + \text{Na}_2\text{SO}_4(\text{aq}) \rightarrow \text{BaSO}_4(\text{s}) + 2\text{NaCl}(\text{aq})$
(c) $\text{CaCO}_3(\text{s}) \rightarrow \text{CaO(s)} + \text{CO}_2(\text{g})$
(d) $\text{C(s)} + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g})$
21. The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula:
- (a) XY (b) XY_2
(c) XY_3 (d) X_3Y_3
22. The atomic radius (pm) of Li, Na, K and Rb varies in the order:
- (a) $\text{Na} < \text{K} < \text{Rb} < \text{Li}$ (b) $\text{K} < \text{Na} < \text{Li} < \text{Rb}$
(c) $\text{Li} < \text{Na} < \text{K} < \text{Rb}$ (d) $\text{Rb} < \text{K} < \text{Na} < \text{Li}$
23. The electron affinity value (kJ mol^{-1}) of fluorine (F) is less than:
- (a) Hydrogen (H) (b) Lithium (Li)
(c) Oxygen (O) (d) Chlorine (Cl)
24. Which of the following is an example of strong electrolyte?
- (a) H_2CO_3 (b) NH_4OH
(c) NaCl (d) $\text{COOH} - \text{COOH}$
25. If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be:
- (a) two times (b) five times
(c) ten times (d) twenty times
26. Bleaching powder is manufactured by the reaction of:
- (a) CaCl_2 and CaCO_3 (b) Cl_2 and Ca(OH)_2
27. (c) Cl_2 and CaSO_4 (d) Cl_2 and $\text{Ca(HCO}_3)_2$
The hard glass is obtained by fusing:
- (a) soda ash, sand and limestone
(b) a mixture of sand, lime, borax and alkali carbonates
(c) potassium carbonate and limestone
(d) potassium carbonate, lead oxide and sand
28. Which of the following metals is most reactive?
- (a) Aluminium (b) lead
(c) Mercury (d) Silver
29. Which of the following alloys contains chromium?
- (a) Steel (b) Stainless steel
(c) Magnadium (d) Brass
30. Which of the following is monomer of natural rubber?
- (a) Chloroethene (b) Chloroprene
(c) Isoprene (d) Buta-1, 3-diene
31. Alkaline KMnO_4 oxidises propanone to:
- (a) propanoic acid (b) ethanoic acid
(c) methanoic acid (d) oxalic acid
32. Which of the following metals can displace Zn from ZnSO_4 solution?
- (a) Calcium (b) Copper
(c) Iron (d) Mercury
33. Heating of sodium ethanoate with soda lime yields:
- (a) ethane (b) methane
(c) ethanol (d) methanol
34. The compound formed by the reaction of ethyne with bromine is:
- (a) $\text{Br}-\text{CH}=\text{CH}-\text{Br}$ (b) $\text{Br}-\text{CH}_2-\text{CHBr}_2$
(c) $\text{CH}_2=\text{CH}-\text{Br}$ (d) $\text{Br}_2\text{CH}-\text{CHBr}_2$
35. Cell were first discovered by Robert Hooke in the year:
- (a) 1665 (b) 1674 (c) 1831 (d) 1839
36. The plant cells have a rigid cell wall that lies:
- (a) Outside the plasma membrane
(b) Inside the plasma membrane
(c) In between the plasma membranes
(d) None of the above
37. During mitosis nucleolus and nuclear membrane are lost in stage:
- (a) Prophase (b) Metaphase
(c) Anaphase (d) Telophase
38. An example of simple and permanent tissue is:
- (a) Xylem (b) Phloem
(c) Sclerenchyma (d) All the above
39. The shape of squamous epithelial tissue is:
- (a) Cubical (b) Flattened
(c) Pillar like (d) None of the above
40. Fresh water sponge belongs to phylum:
- (a) Arthropoda (b) Annelida

13 Maths Entrance Exam Papers

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French for Common Entrance 13+ Exam Practice Questions and Answers (for the June 2022 exams) Nigel Pearce, Joyce Capek, 2019-10-28 Please note this resource is suitable for the exams up to June 2022 New revision resources will be available from Spring 2022 for the exams from November 2022 Exam board ISEB Level 13 Subject French First exam Autumn 2019 This book features a wealth of exam style questions which have been completely revised and updated for the new French ISEB Common Entrance 13 exam beginning Autumn 2019 Exercises are differentiated by Levels 1 and 2 and are modelled on ISEB papers covering a variety of topics This book also includes answers to every question as well as marking guidance Endorsed by ISEB for the new exam beginning Autumn 2019 Provides comprehensive practice for French at 13 Common Entrance Builds students confidence and helps identify any gaps in knowledge The accompanying audio for this book is available free from galorepark.co.uk frenchaudio

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Table of Contents 13 Maths Entrance Exam Papers

1. Understanding the eBook 13 Maths Entrance Exam Papers
 - The Rise of Digital Reading 13 Maths Entrance Exam Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying 13 Maths Entrance Exam Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 13 Maths Entrance Exam Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 13 Maths Entrance Exam Papers
 - Personalized Recommendations
 - 13 Maths Entrance Exam Papers User Reviews and Ratings
 - 13 Maths Entrance Exam Papers and Bestseller Lists
5. Accessing 13 Maths Entrance Exam Papers Free and Paid eBooks
 - 13 Maths Entrance Exam Papers Public Domain eBooks
 - 13 Maths Entrance Exam Papers eBook Subscription Services
 - 13 Maths Entrance Exam Papers Budget-Friendly Options

6. Navigating 13 Maths Entrance Exam Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - 13 Maths Entrance Exam Papers Compatibility with Devices
 - 13 Maths Entrance Exam Papers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 13 Maths Entrance Exam Papers
 - Highlighting and Note-Taking 13 Maths Entrance Exam Papers
 - Interactive Elements 13 Maths Entrance Exam Papers
8. Staying Engaged with 13 Maths Entrance Exam Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 13 Maths Entrance Exam Papers
9. Balancing eBooks and Physical Books 13 Maths Entrance Exam Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 13 Maths Entrance Exam Papers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 13 Maths Entrance Exam Papers
 - Setting Reading Goals 13 Maths Entrance Exam Papers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 13 Maths Entrance Exam Papers
 - Fact-Checking eBook Content of 13 Maths Entrance Exam Papers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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