

|                            | Covalent Bonds                                                                                                                                                                                                                                    | Ionic Bonds                                                                                                                                                                                                                                    |
|----------------------------|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| State at room temperature: | gases, liquids or low-melting solids                                                                                                                                                                                                              | Crystalline solids                                                                                                                                                                                                                             |
| Polarity:                  | Low                                                                                                                                                                                                                                               | High                                                                                                                                                                                                                                           |
| Solubility                 | Few are soluble in water; Many are soluble in organic liquids                                                                                                                                                                                     | Many are water soluble; Not soluble in organic liquids                                                                                                                                                                                         |
| Formation:                 | A covalent bond is formed between two non-metals that have similar electronegativities. Neither atom is "strong" enough to attract electrons from the other. For stabilization, they share their electrons from outer molecular orbit with others | An ionic bond is formed between a metal and a non-metal. Non-metals(-ve ion) are "stronger" than the metal(+ve ion) and can get electrons very easily from the metal. These two opposite ions attract each other and form the ionic bond.      |
| Conductivity:              | Do not conduct electricity                                                                                                                                                                                                                        | Conduct electricity when molten or dissolved in water                                                                                                                                                                                          |
| Melting point:             | Low                                                                                                                                                                                                                                               | High                                                                                                                                                                                                                                           |
| What is it?:               | Covalent bonding is a form of chemical bonding between two non metallic atoms which is characterized by the sharing of pairs of electrons between atoms and other covalent bonds.                                                                 | Ionic bond, also known as electrovalent bond, is a type of bond formed from the electrostatic attraction between oppositely charged ions in a chemical compound. These kinds of bonds occur mainly between a metallic and a non metallic atom. |
| Boiling point:             | Low                                                                                                                                                                                                                                               | High                                                                                                                                                                                                                                           |
| Examples:                  | Methane ( $\text{CH}_4$ ), Hydrochloric acid ( $\text{HCl}$ )                                                                                                                                                                                     | Sodium chloride ( $\text{NaCl}$ ), Sulfuric Acid ( $\text{H}_2\text{SO}_4$ )                                                                                                                                                                   |
| Occurs between:            | Two non-metals or a non-metal and a metalloid                                                                                                                                                                                                     | One metal and one non-metal                                                                                                                                                                                                                    |

# Reinforcement Section 2 Types Of Bonds Answers

**Wolfgang Guggemos**

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